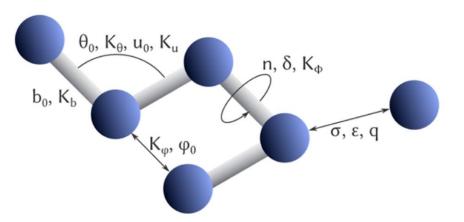
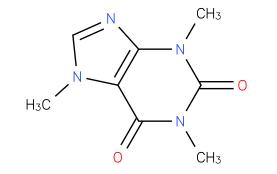


# **Classical Force Fields**

#### Parameterisation

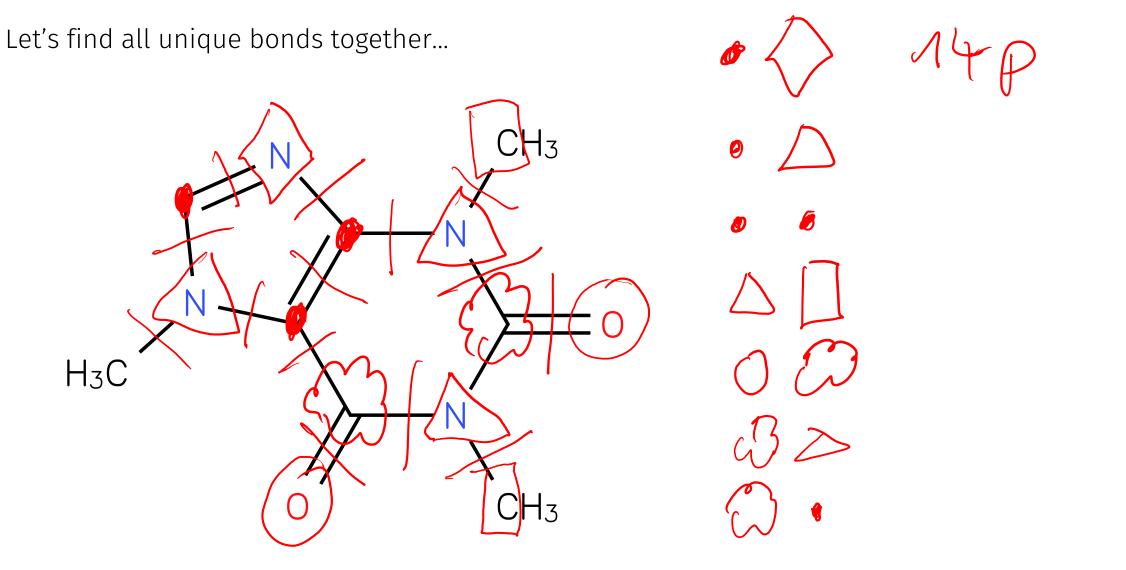
$$E = \sum_{\text{bonds}} K_b (b - b_0)^2 + \sum_{\text{angles}} K_\theta (\theta - \theta_0)^2$$
  
+ 
$$\sum_{\text{dihedrals}} K_\phi (1 + \cos(n\phi - \delta))$$
  
+ 
$$\sum_{\text{improper}} K_\phi (\phi - \phi_0)^2 + \sum_{\text{Urey-Bradley}} K_u (u - u_0)^2$$
  
+ 
$$\sum_{i < j} 4\varepsilon \left[ \left( \frac{\sigma_{ij}}{r_{ij}} \right)^{12} - \left( \frac{\sigma_{ij}}{r_{ij}} \right)^6 \right] + \sum_{i < j} \frac{q_i q_j}{4\pi \varepsilon_0 r_{ij}}$$





- Elements become *atom types*
- Let's find all unique bonds together...

#### Parameterisation

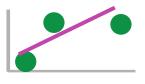


... + angles, + dihedrals

#### Parametrisation: Challenges

- Fixed, parametric form

High bias, sub-selects chemical space



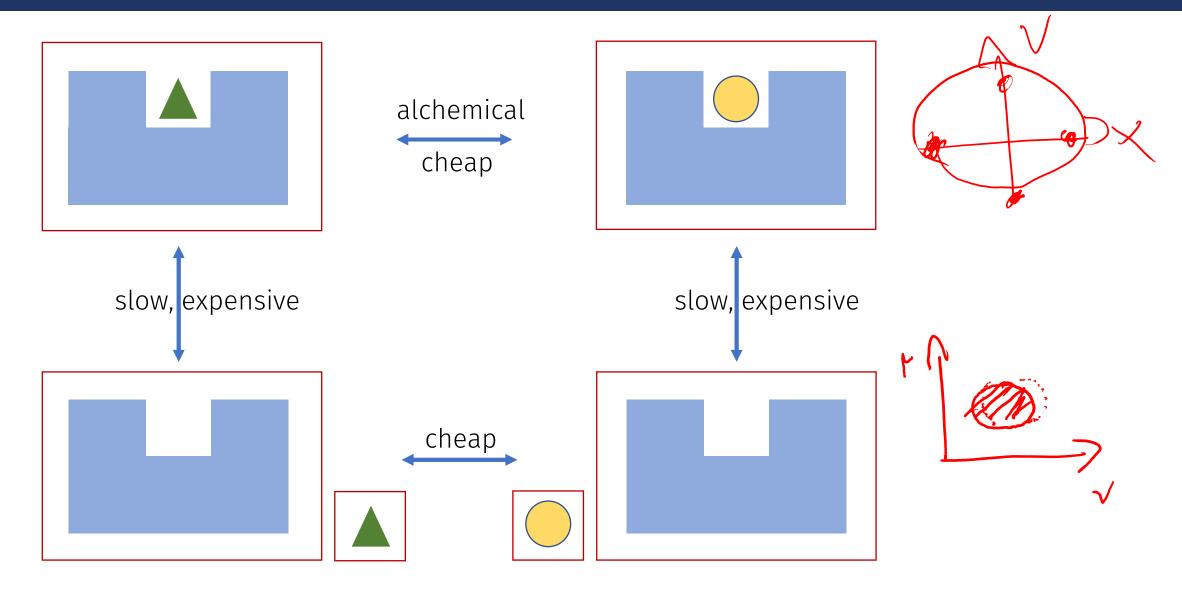
- Atom typing

Non-unique, rarely fully automatic

- e.g. sp2, sp3 carbons but how to detect them? It's a (vague) quantum definition...
- New compound, new parametrization Restricts curiosity
  - More of the same: easy, try something new: hard
- Manually but assisted fitting against
  - DFT Hard to match geometries, hardly possible to get ensembles
  - Experiment Rarely possible to match both ensemble and geometries

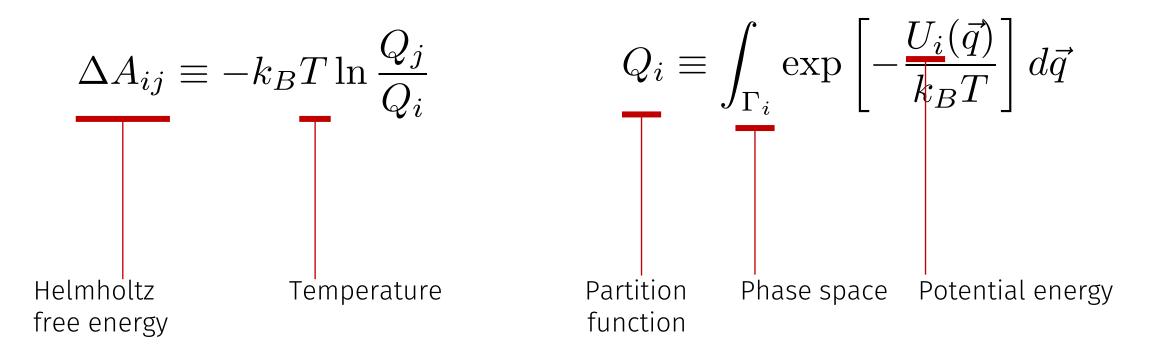
Classical Use Case: Alchemical changes

#### Alchemical changes



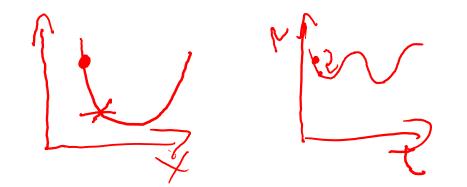
## Theory

- Two states: i, j
- NVT ensemble, equilibrated



### Theory: Molecular dynamics

- Need: ensemble average
- Ergodicity: Time average = Ensemble average
- Propagate in small steps (~fs) explicit positions from repeated force evaluations
- Challenges
  - Numerically stable
  - Time reversibility
  - Thermostats / Barostats
  - Equilibration



Questions:

- Why are classical calculations typically time-reversible but quantum mechanical calculations are not?
- What does it mean if a setup is lacking time-reversibility?