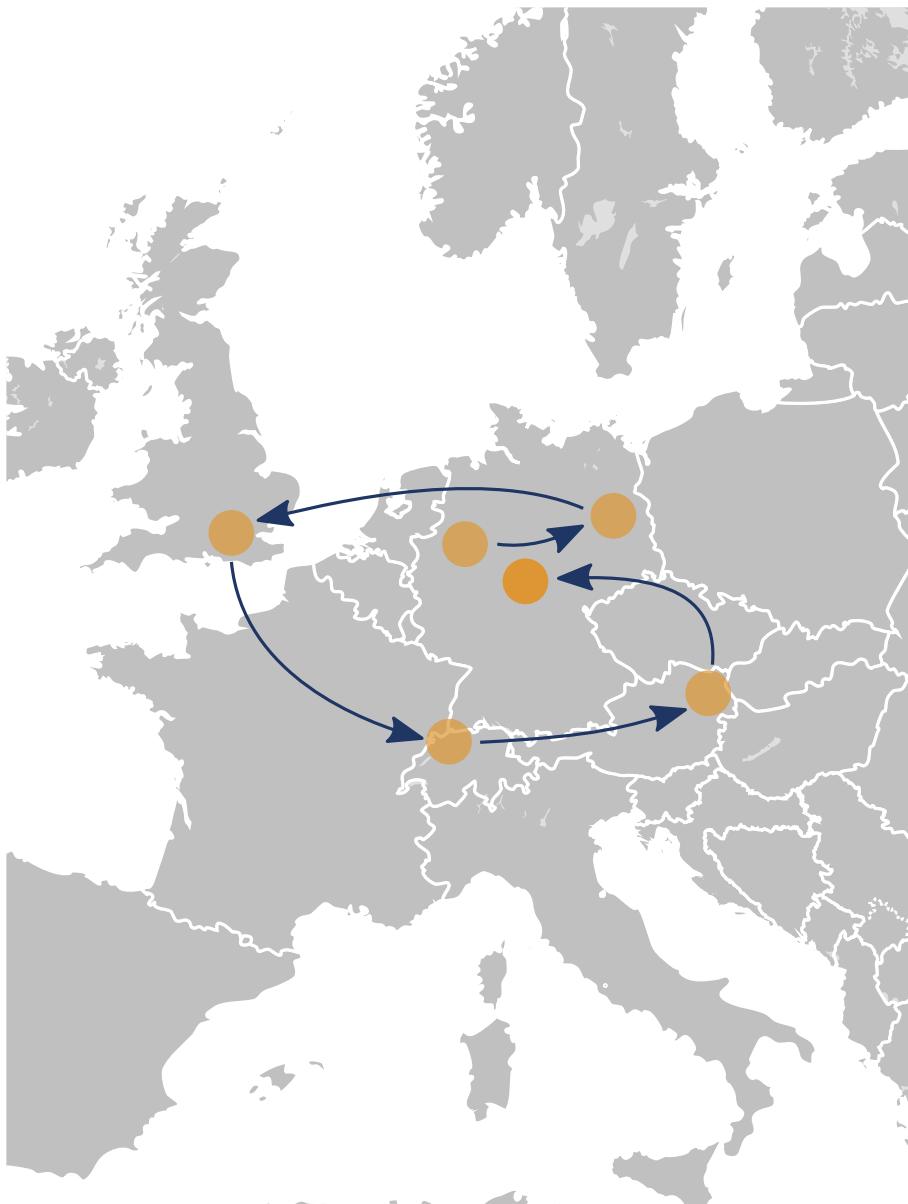


Computational Material Design and the Curse of Dimensionality

Guido Falk von Rudorff, University of Kassel

About me

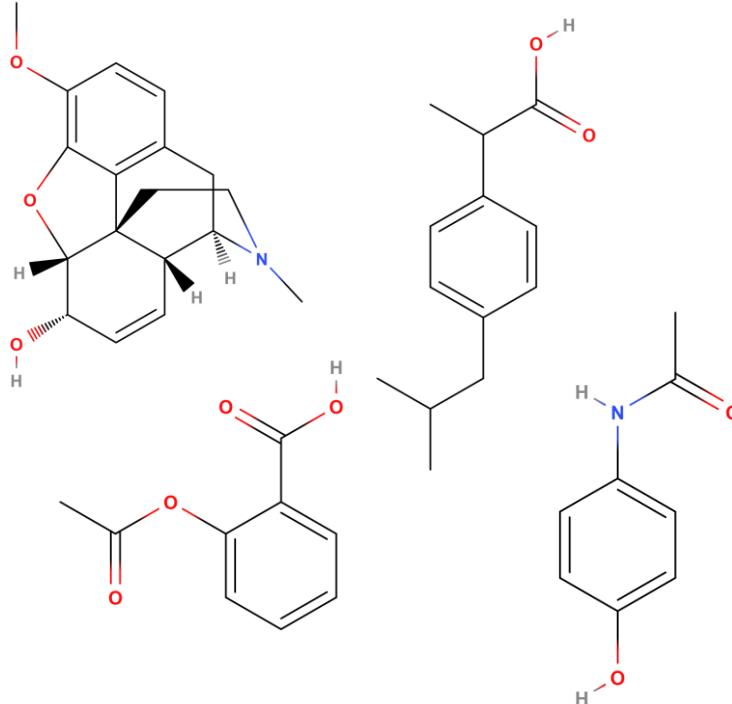
2



U N I K A S S E L
V E R S I T Ä T

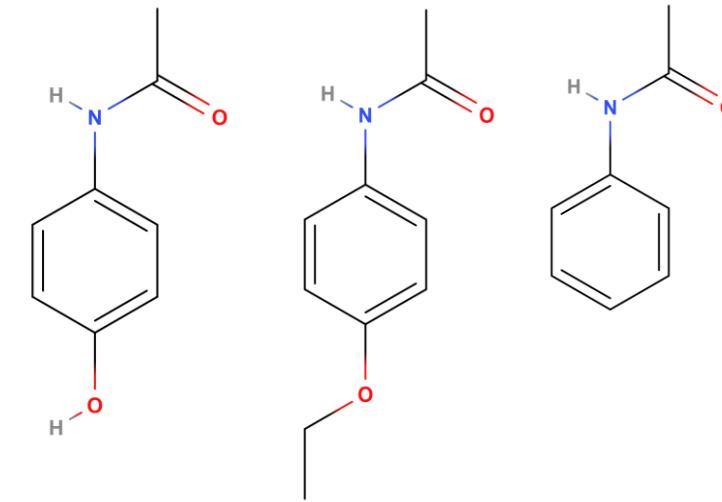
Introduction

3



Global Search Problem

Which class of compounds?

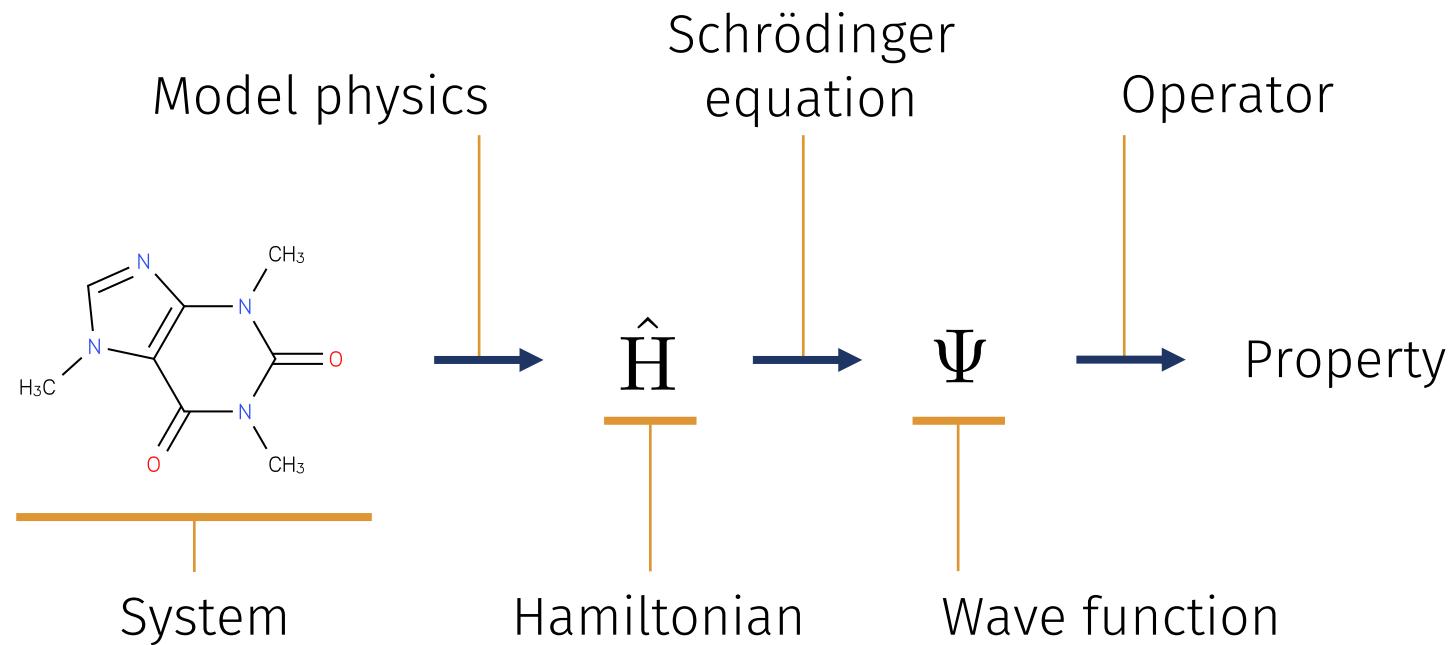


Local Search Problem

Which particular species within that class?

Solved?

4



Wave Function

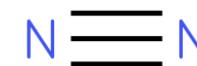
5

$$\Psi = \Psi(\mathbf{r}_1, \mathbf{r}_2, \mathbf{r}_3, \dots, \mathbf{r}_n)$$

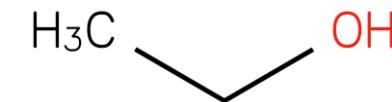
Methane



N₂



Ethanol



Solved by approximations in computational chemistry?



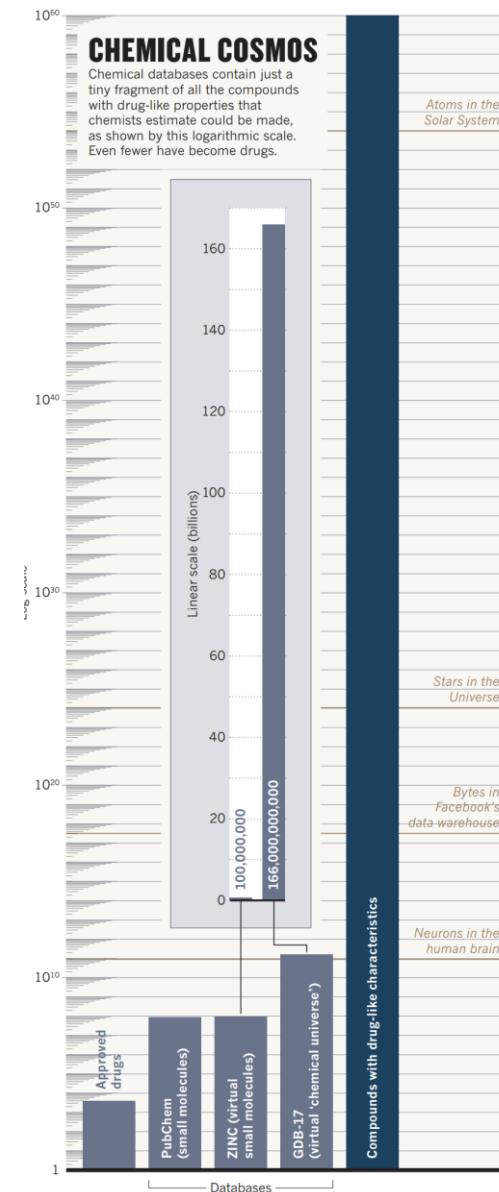
Scaling of Molecules

Commercial databases

- 164 million molecules
- 15k added daily

Scale

- One person: 1 million compounds/second
- 10 billion people on earth
- 10^{26} universe ages to go through



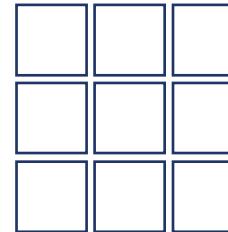
Scaling of Materials

7

Face centered cubic and 70 elements only

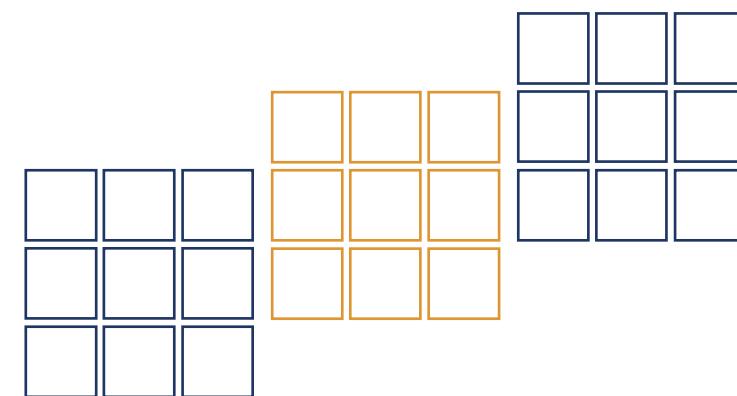
9 primitive cells

- Binary 10^7
- Ternary: 10^{13}
- Quaternary: 10^{15}



27 primitive cells

- Binary: $\sim 10^{17}$
- Ternary: $\sim 10^{29}$
- Quaternary: $\sim 10^{36}$



3D Geometries

8

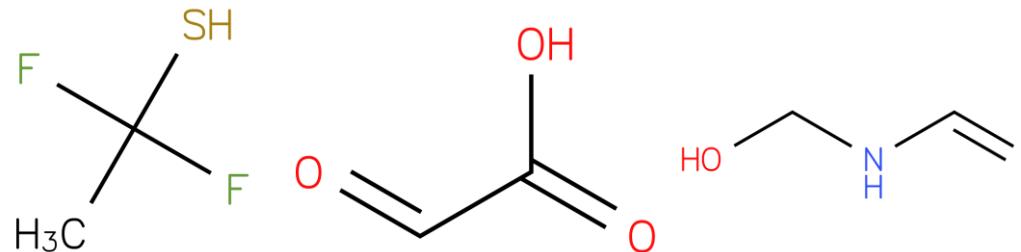
Take some atoms from {C, O, N, F, S}, H-saturated



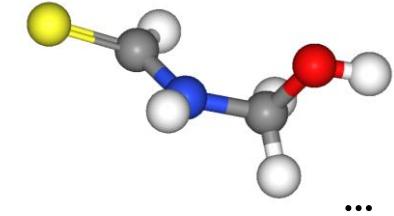
349:

$\text{C}_2\text{H}_4\text{F}_2\text{S}$
 $\text{C}_2\text{H}_2\text{O}_3$
 $\text{C}_2\text{H}_5\text{NOS}$
...

9,917:



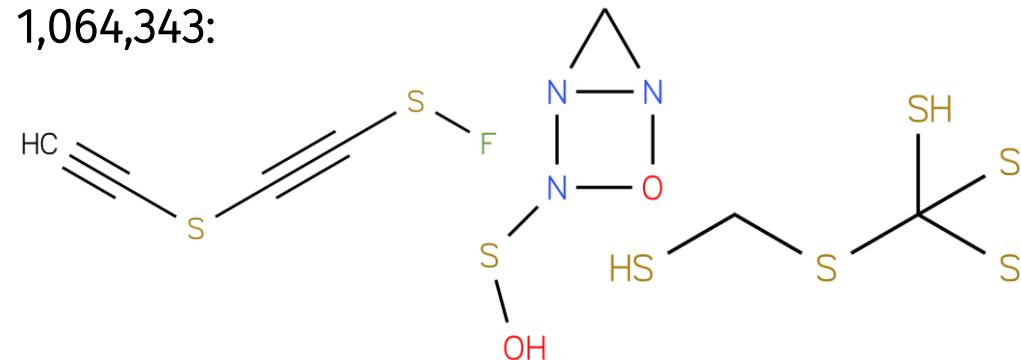
~52k:



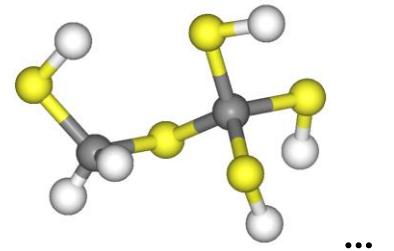
1,050:

C_4HFS_2
 $\text{CH}_3\text{N}_3\text{O}_2\text{S}$
 $\text{C}_2\text{H}_6\text{S}_5$
...

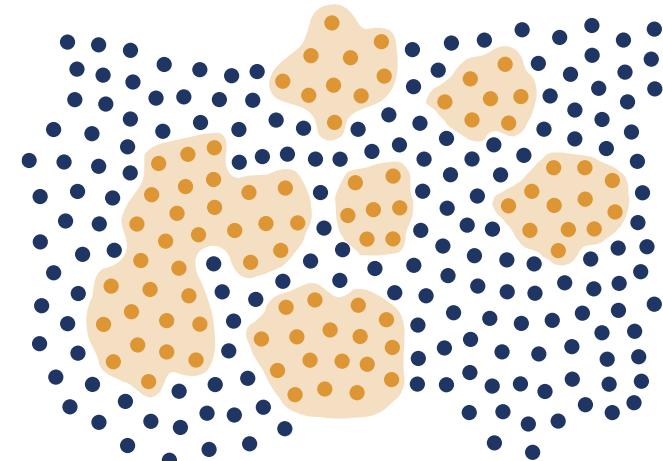
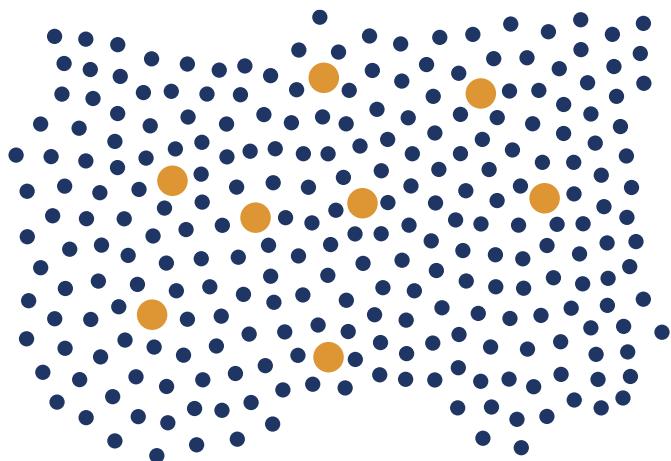
1,064,343:



~23M:



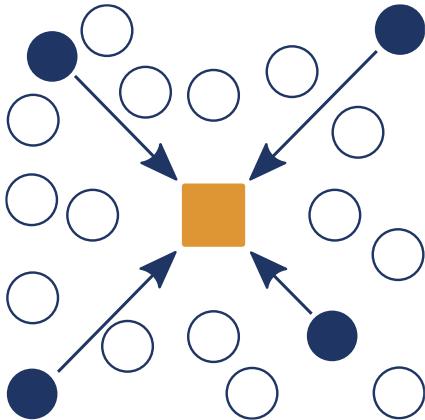
Speed does not matter:
even enumeration is impossible.



Approaches

10

Machine Learning



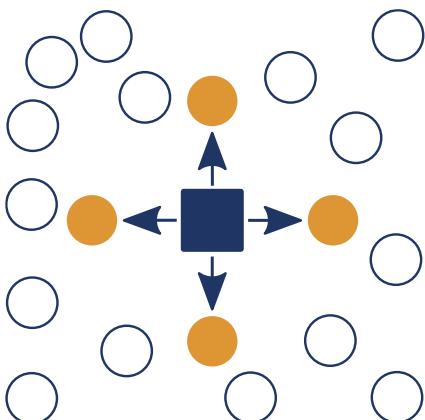
Foundations | Statistical modelling

Accuracy | Systematically improvable through data and training

Specialty | Universal, scale-bridging, data-driven approach

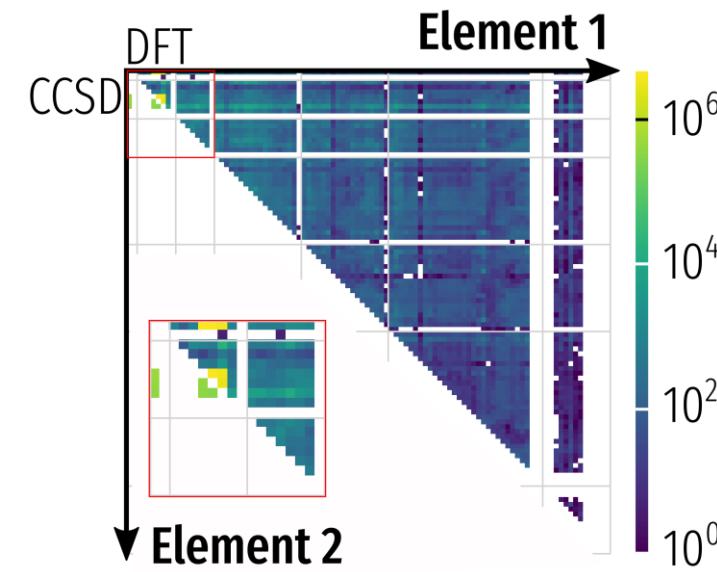
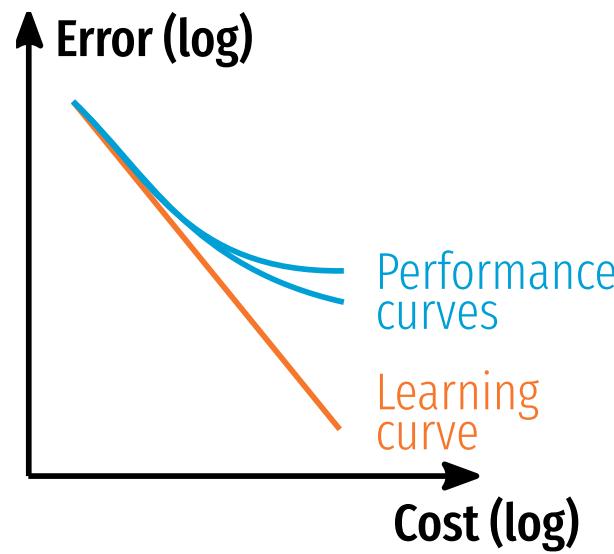
Limitation | Requires training data, no black box

Quantum Alchemy



Data Availability

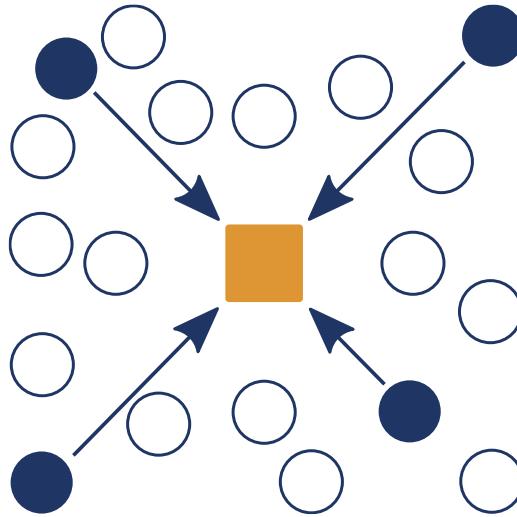
11



Low-Data Regime

12

Machine Learning



Kernel-Ridge-Regression

- Efficient in the low-data regime (around 1k points)
- Ingredients
 - Representation
 - Similarity measure
 - Observed properties
- Training
 - Pairwise similarities
 - Model coefficients
- Predictions
 - Compare to training

$$\mathbf{M}$$

$$k(\mathbf{M}_i, \mathbf{M}_j)$$

$$\mathbf{y}$$

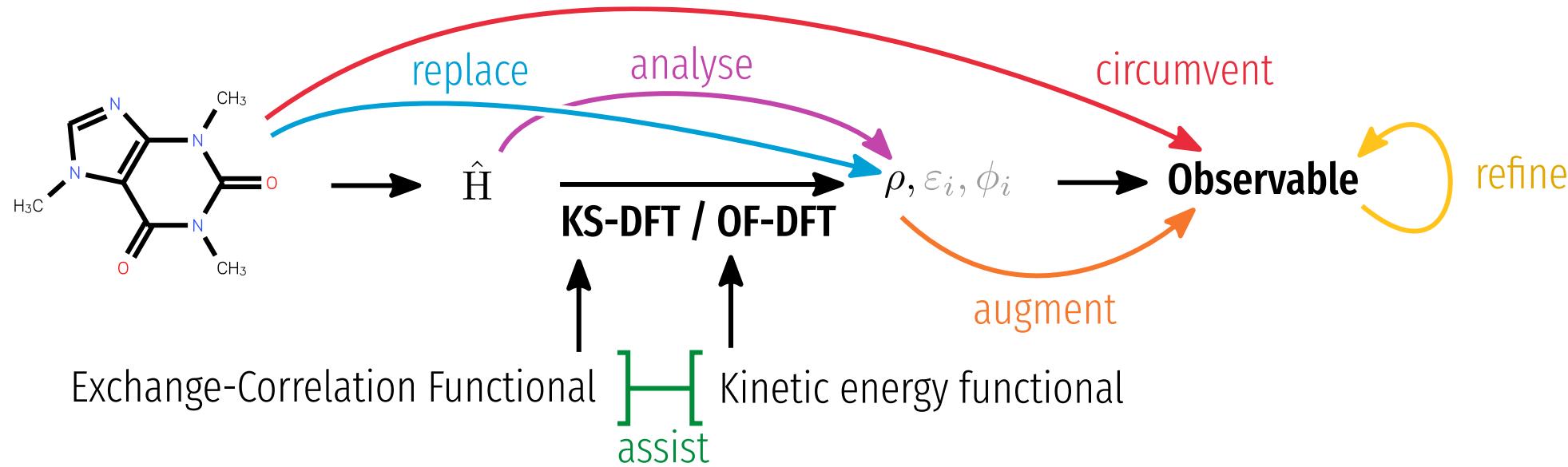
$$\mathbf{K}$$

$$\boldsymbol{\alpha} = (\mathbf{K} + \lambda \mathbf{I})^{-1} \mathbf{y}$$

$$\tilde{q}(\mathbf{M}) = \sum_i \alpha_i k(\mathbf{M}, \mathbf{M}_i)$$

Strategies

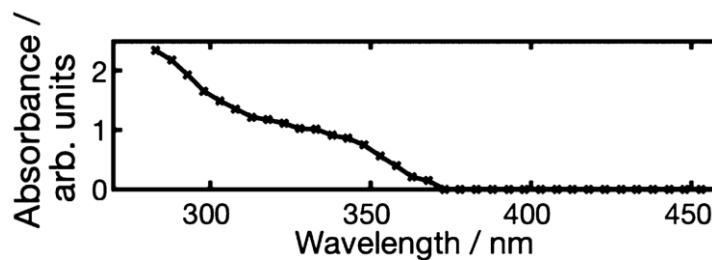
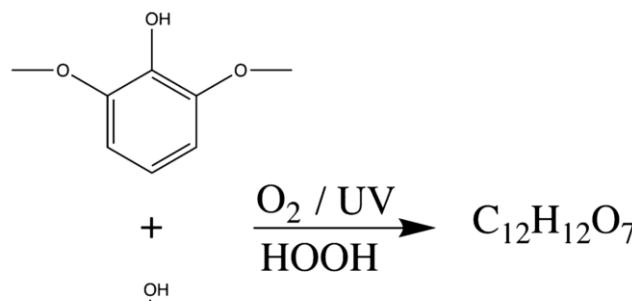
13



Filter by Spectrum

14

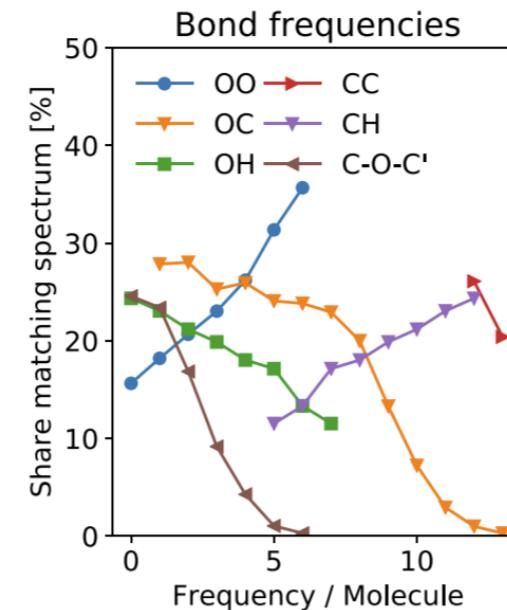
Experiment



Search space

Molecular graphs: 264 M

Stable molecules: 123 M



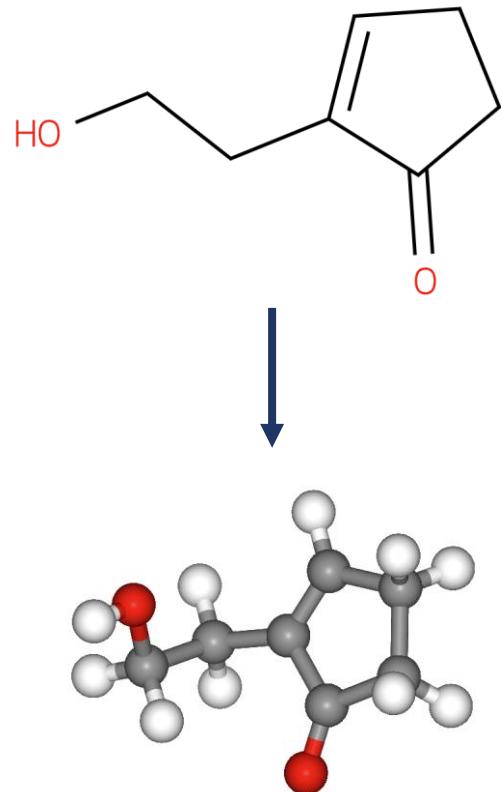
Millions at once!

Guide experiment

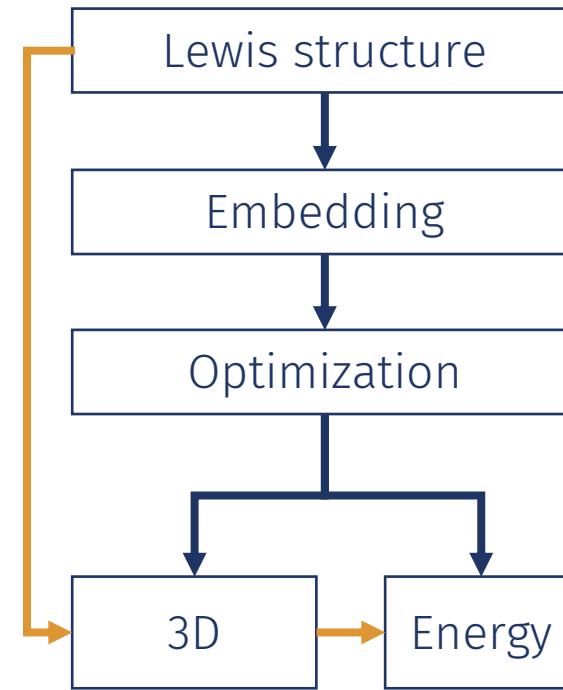
How many molecules are left and which feature to measure next?

Filter by Geometry

15



Traditional / G2S



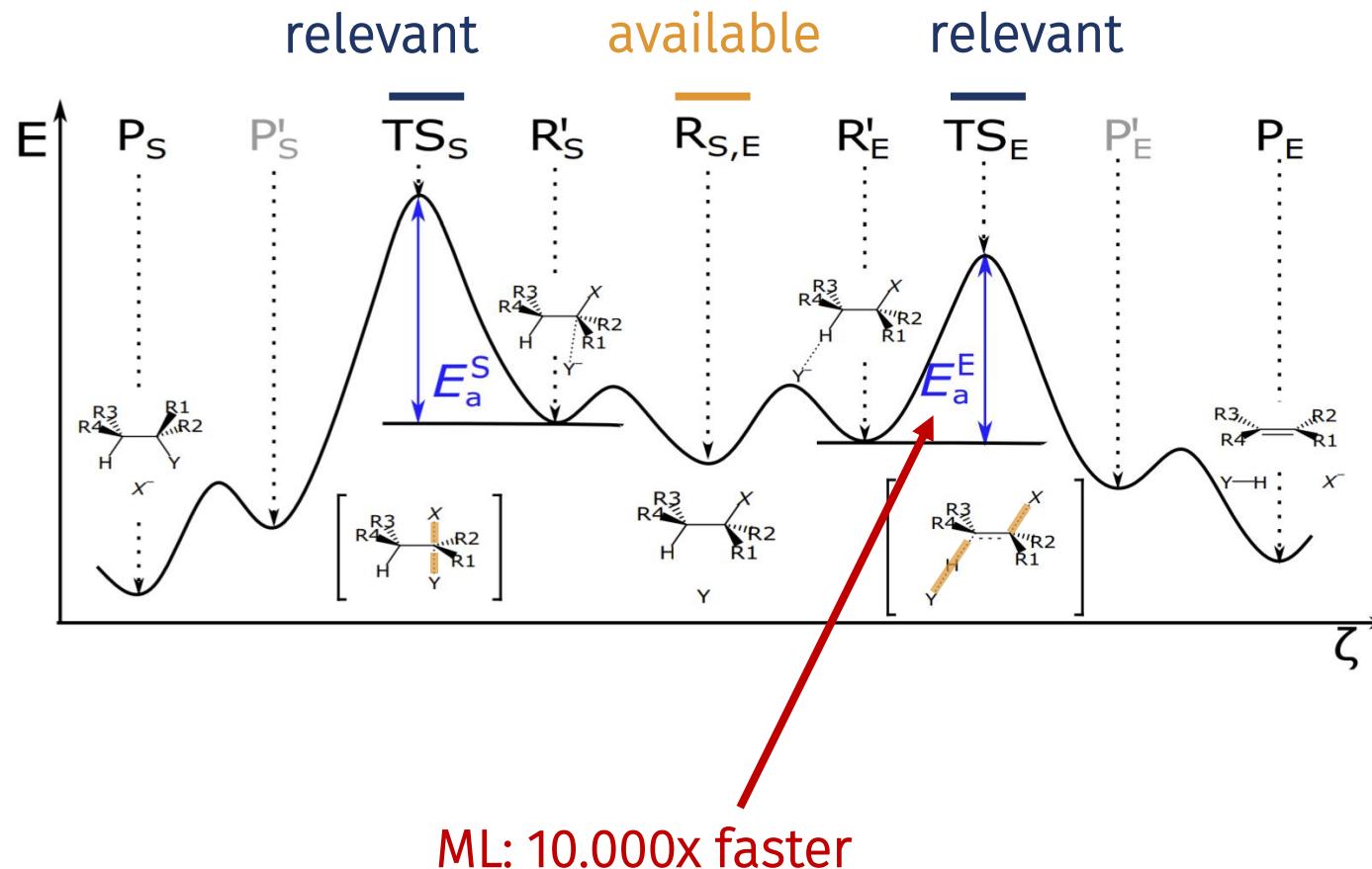
G2S

- Closer to DFT than common methods
 - Small molecules
- Applicable to complex chemical spaces
 - Transition state geometries
 - Carbenes
 - Elpasolite crystals

ML: 100.000x faster,
can filter graphs

Filter by Barriers

16

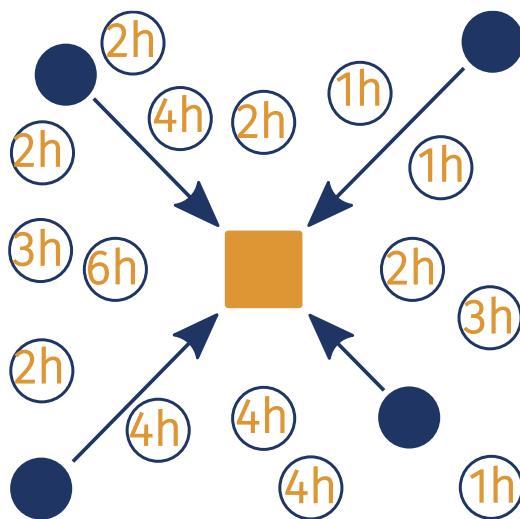
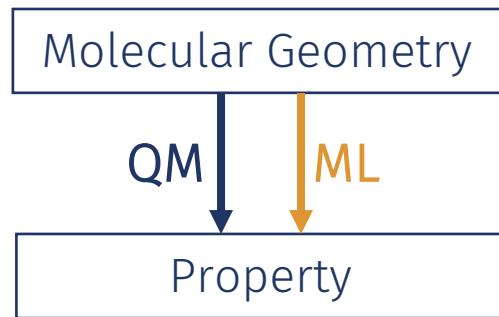


Competing reactions: E2, S_N2

- 4.5k reactions in one new dataset
- Learning activation energies from reactants only reaching 2.5 kcal/mol with 800 data points
- Learning geometries of transition states
 - direct
0.05 Angstrom for distances
 - G2S
0.45 Angstrom heavy-atom RMSD

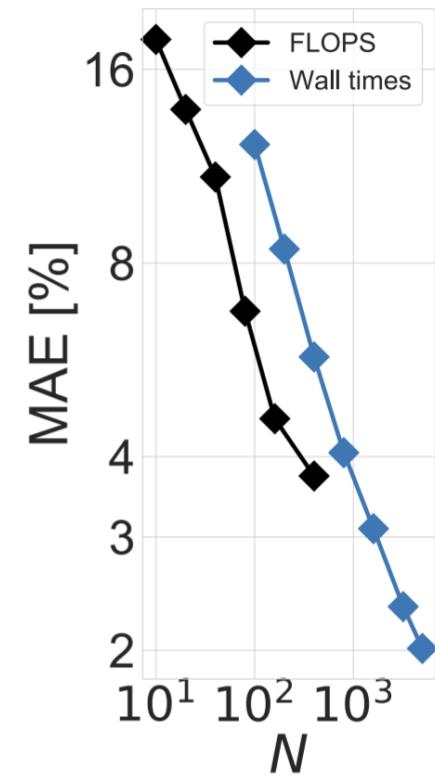
Filter by Cost

17



Computational effort as molecular property

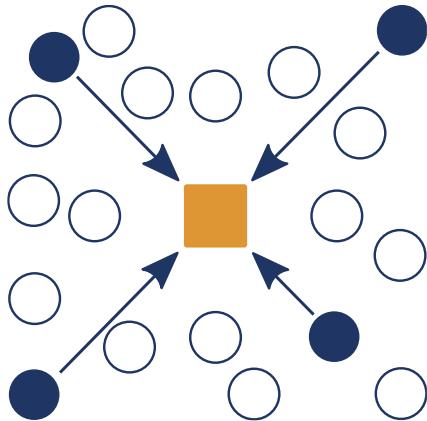
- Improves models
 - Accuracy depends on problem
 - Single points: 2%
 - Transition state search: 25%
 - Geometry optimisations: 40%



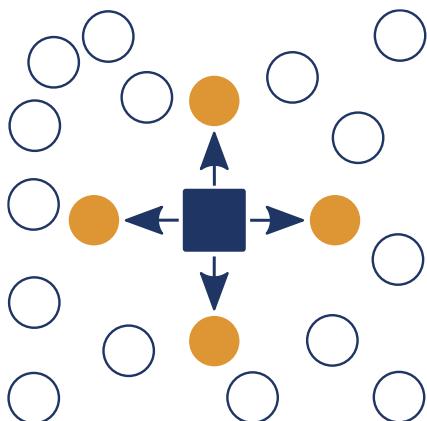
Approaches

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Machine Learning



Quantum Alchemy



Foundations | Perturbation theory

Accuracy | Systematically improvable through higher orders terms

Specialty | Combinatorial scaling with chemical diversity

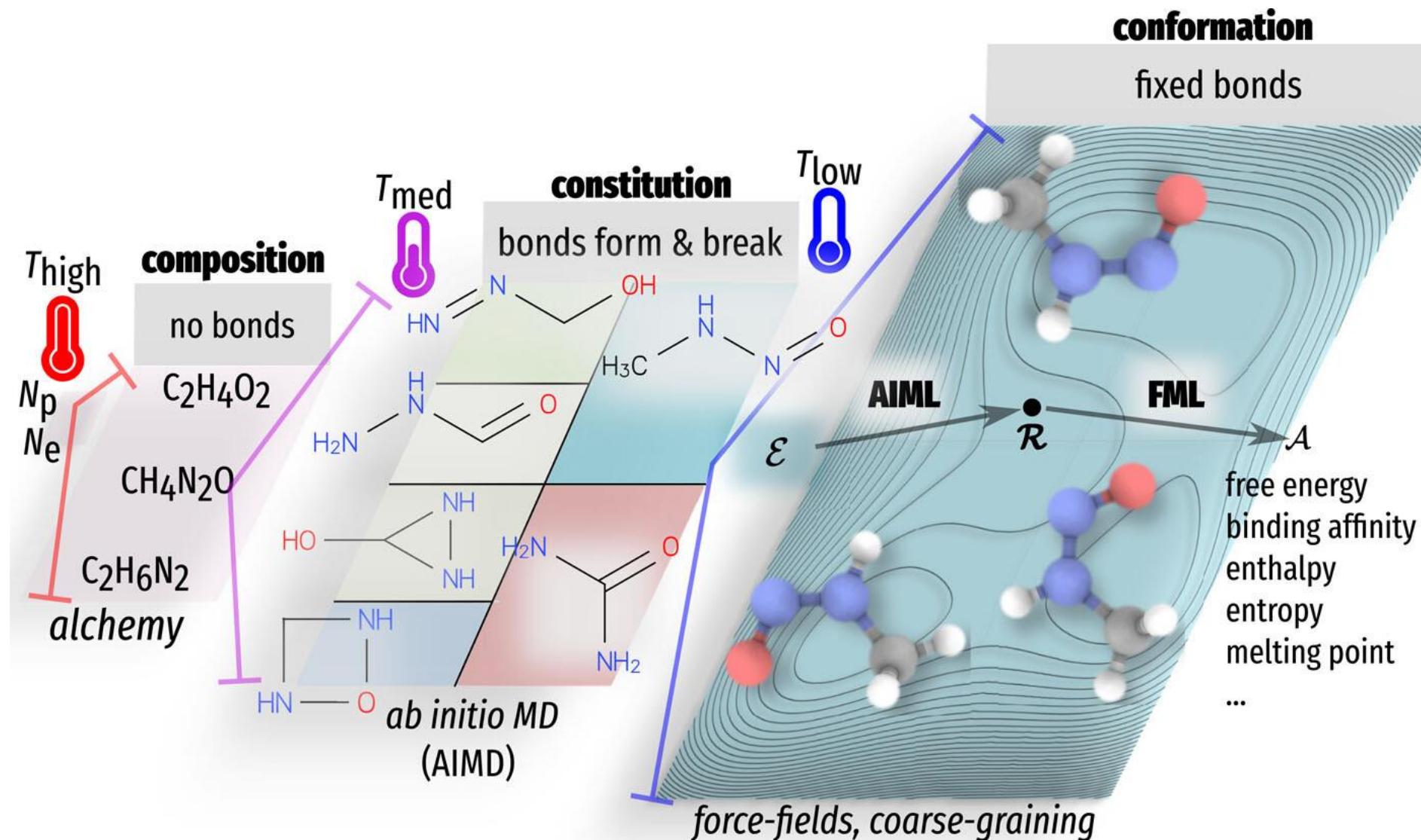
Limitation | Finite range in chemical space



Joseph Wright, 1771

Discrete points?

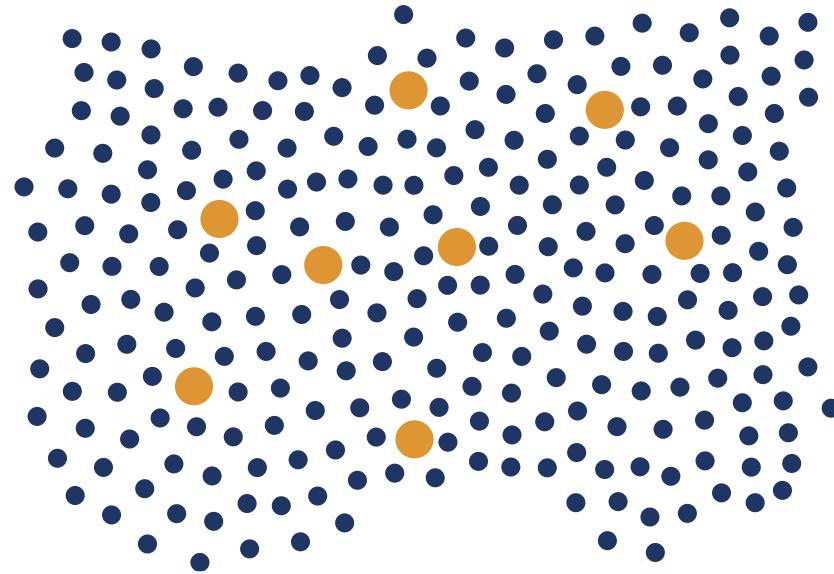
20



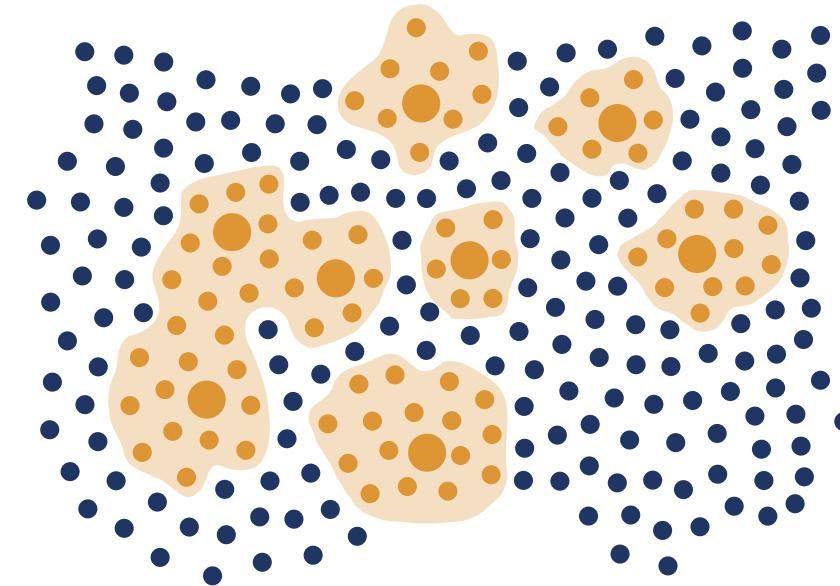
Motivation

21

Without Perturbation



With Perturbation



- Systems/Molecules
- Any
 - Known
 - Approximated

Perspective shift

Few highly accurate calculations
instead of many intermediate ones

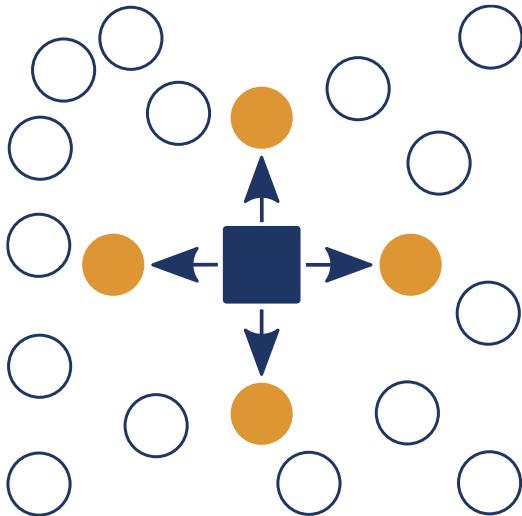
$$\hat{H} = \hat{H}(Z_i, \mathbf{R}_i, N_e, \sigma)$$

4N 1D, close to $\sum_i Z_i$

Quantum Alchemy

23

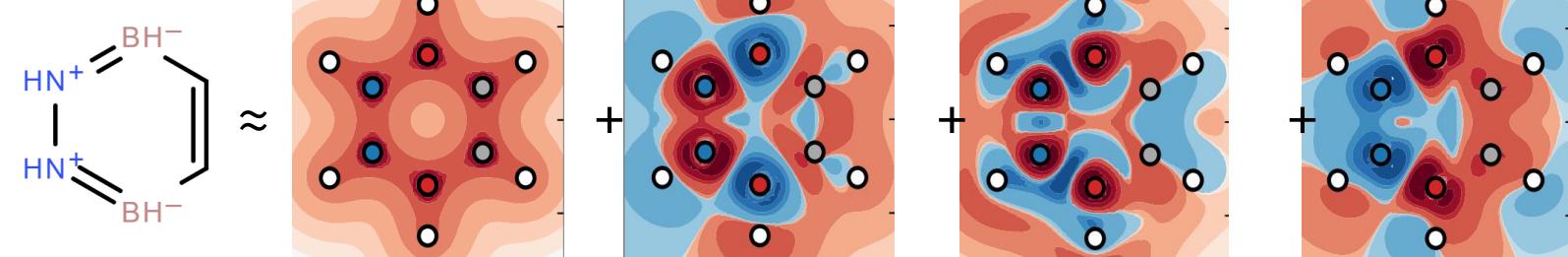
Quantum Alchemy



Taylor expansion

- Energy function of
 - Geometry
 - Nuclear charges
- Idea: obtain dominant leading derivatives, predict many systems

Forces, Vibrations
Alchemical changes



Quantum Alchemy

24

Interpolate between molecular isoelectronic Hamiltonians

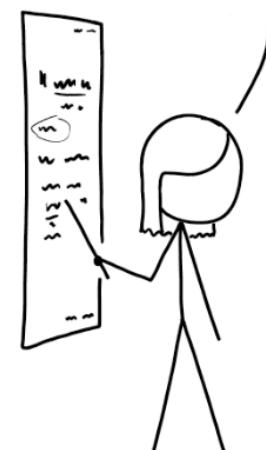
$$\hat{H}(\lambda) \equiv \lambda \hat{H}_t + (1 - \lambda) \hat{H}_r$$

Taylor expansion around reference molecule

$$E_t = E_r + \Delta E^{\text{NN}} + \int_{\Omega} d\mathbf{r} \sum_{n=0}^{\infty} \frac{1}{(n+1)!} \left. \Delta v \frac{\partial^n \rho_{\lambda}(\mathbf{r})}{\partial \lambda^n} \right|_{\lambda=0}$$

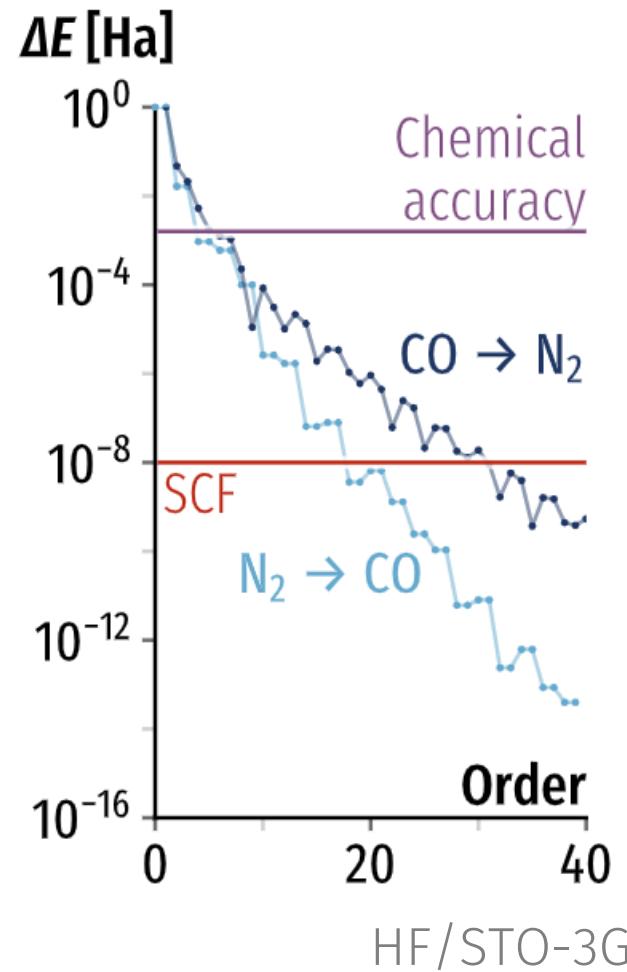
- Gives consistent energies, densities, forces, ...
- Uses the same derivatives for all predictions

AT THIS POINT, YOU'RE PROBABLY
THINKING, "I LOVE THIS EQUATION
AND WISH IT WOULD NEVER END!"
WELL, GOOD NEWS!



Convergence

25

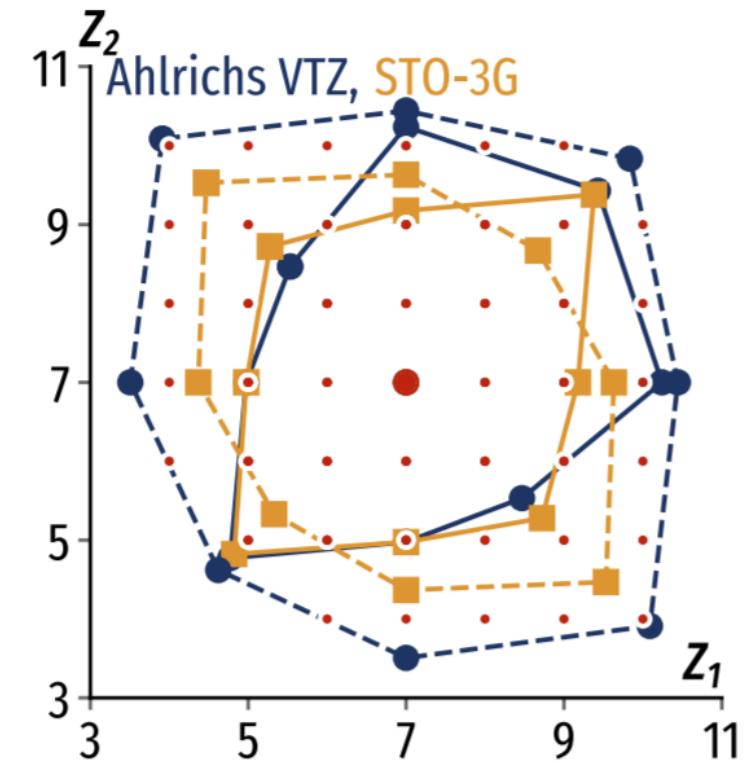


Taylor expansion

- First terms accurate enough
 - Truncate early
- Converges to the right value
- Large convergence radius
- Scales with chemical space

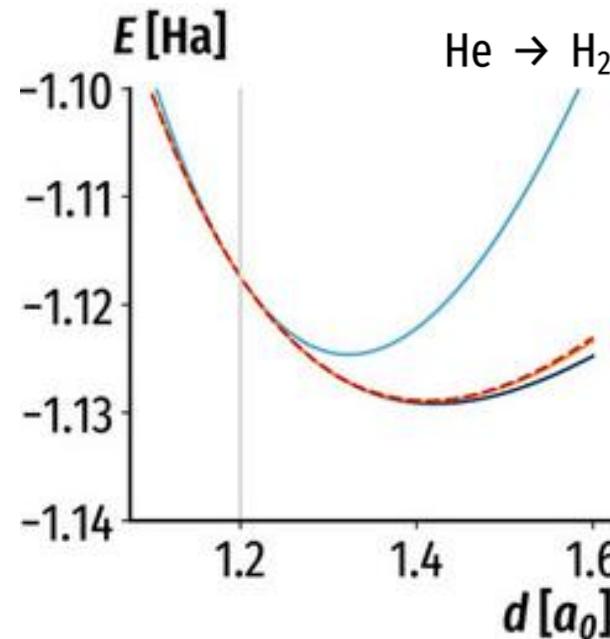
Limits

- Only few orders feasible
- Iterative in chemical space



Geometry Relaxation

26



Taylor expansion

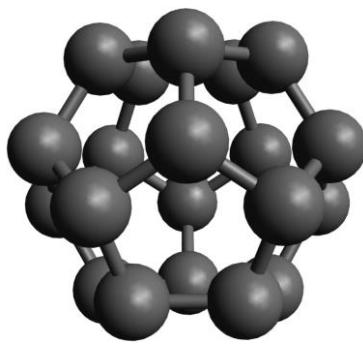
- Large changes still converge (more slowly)
- Geometric response can be recovered

Covalent Interactions

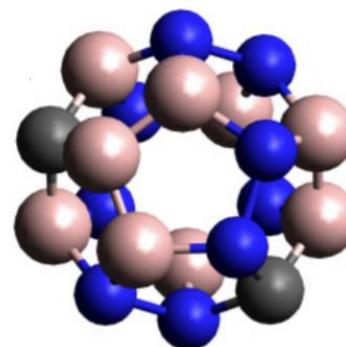
27

Scaling with chemical space

- 1 derivative for second order
- 5 derivatives for third order

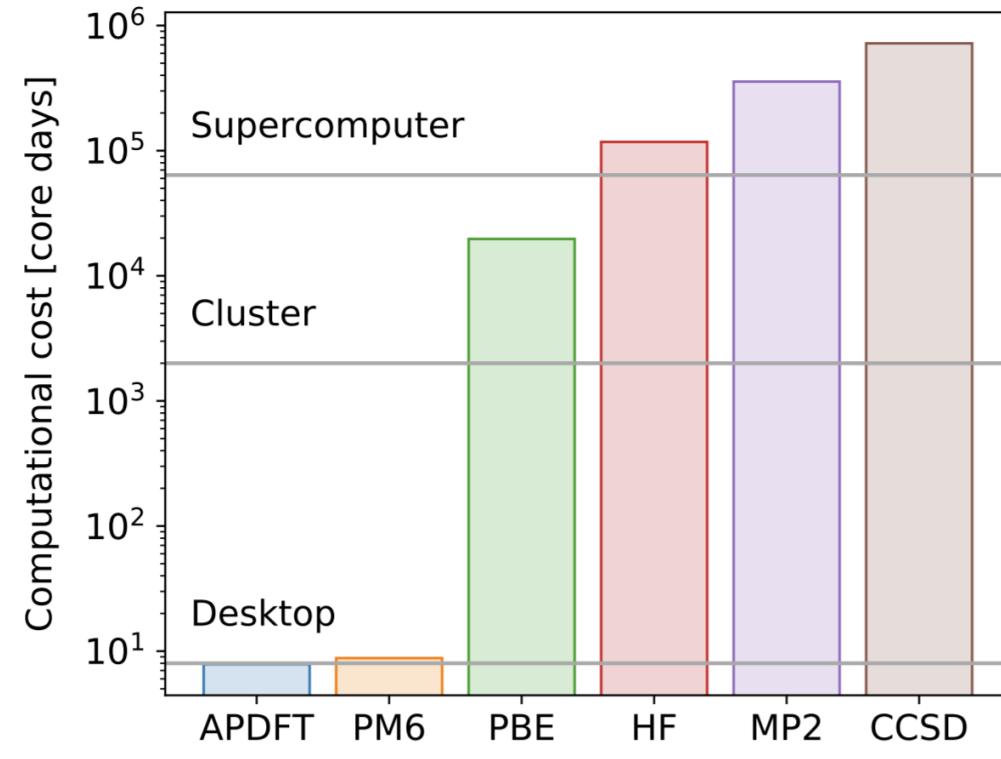


C_{20}



$3.1 \cdot 10^6$
targets

QA: 80.000x faster



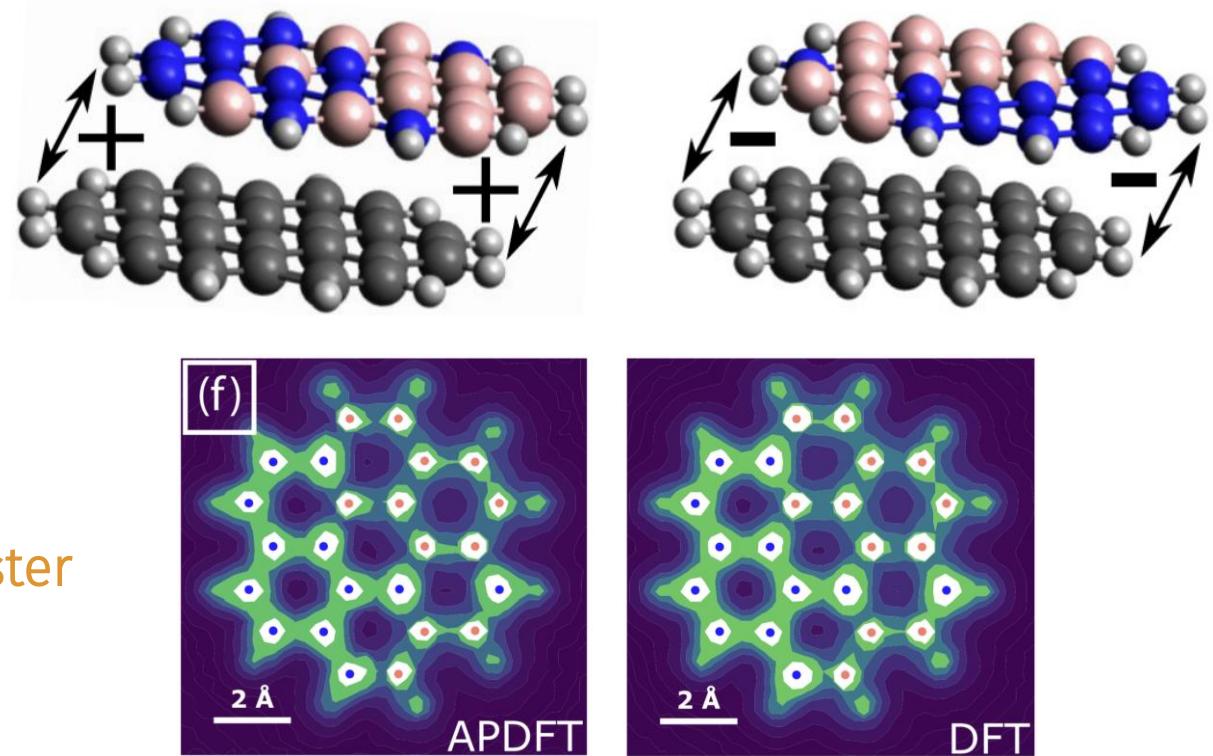
Non-covalent Interactions

28

BN-doped coronene dimer

- Identify most/least attractive doping pattern
- Design case

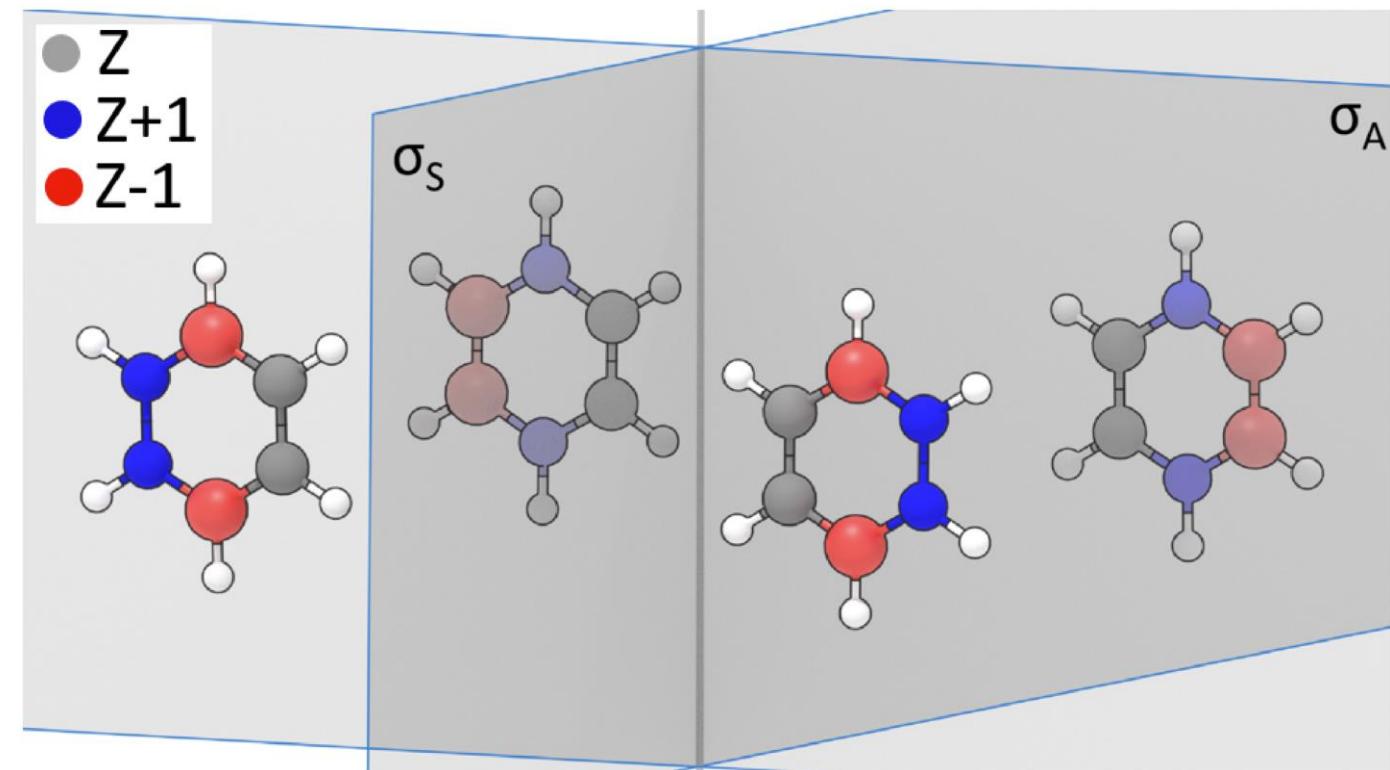
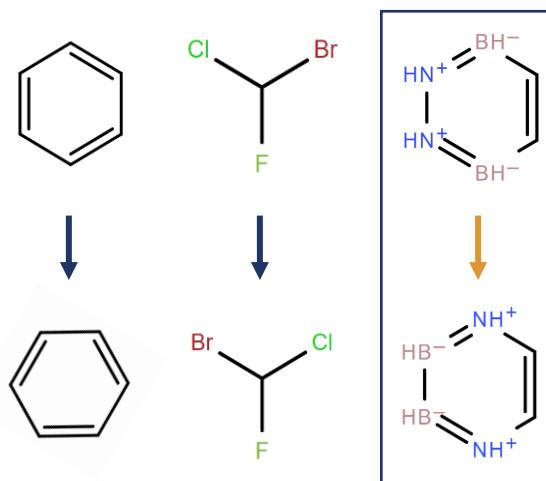
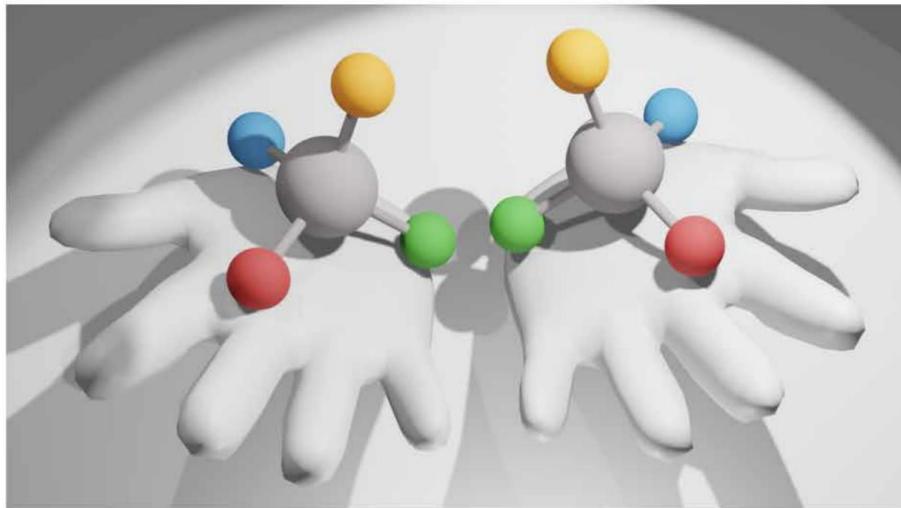
QA: 20.000x faster



$2.8 \cdot 10^{10}$ targets

Alchemical Enantiomers

29

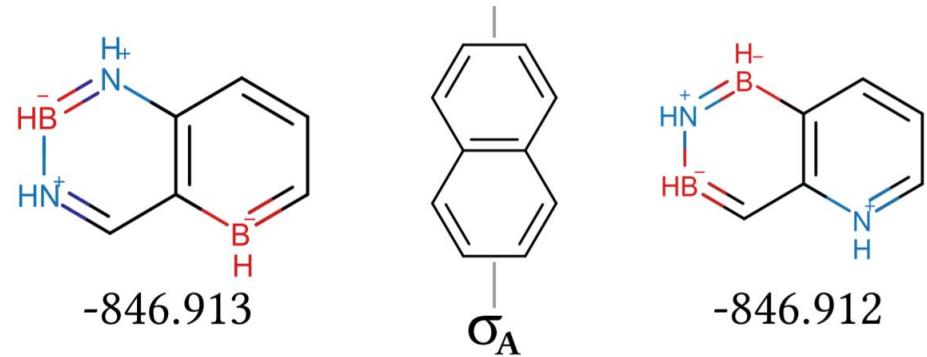


Alchemical Enantiomers

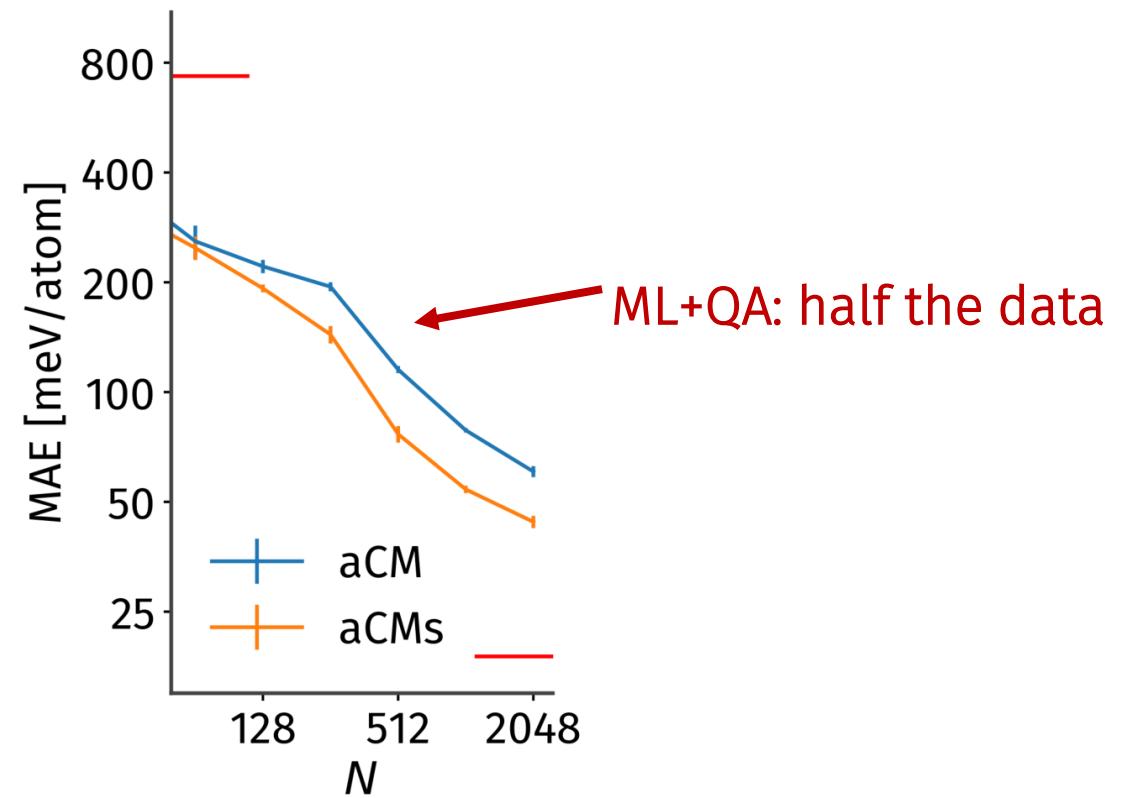
30

Fundamentally new symmetry

Electronic energy only

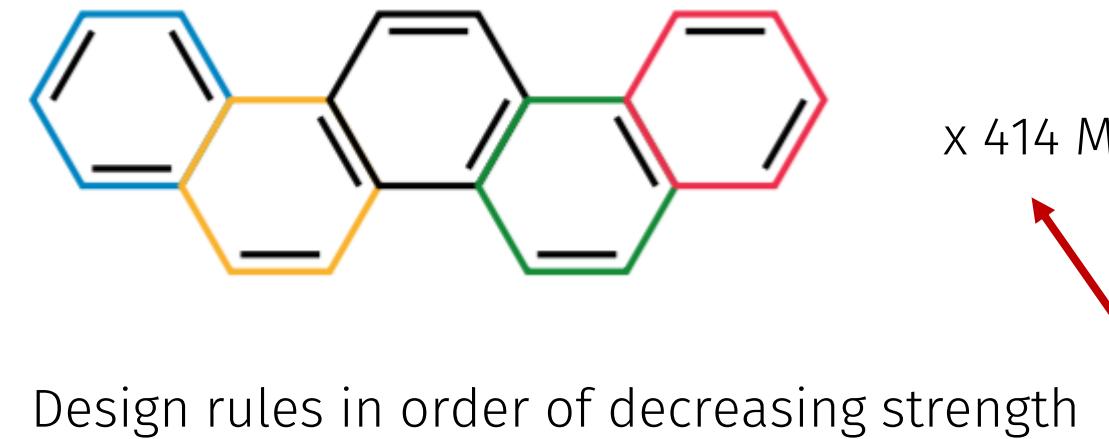
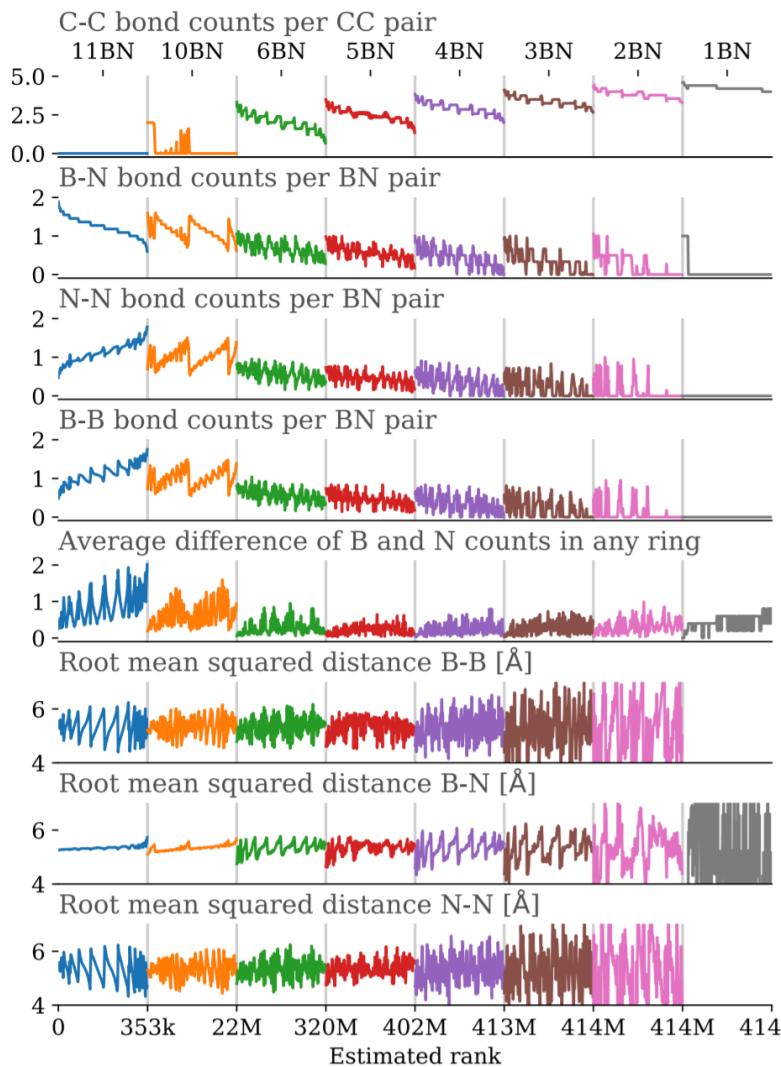


Speed up machine learning



Alchemical Enantiomers

31



Design rules in order of decreasing strength

- Add BN pairs
- Maximize CC bonds
- Substitute sites shared between rings
- Maximize BN bonds
- Avoid N substitutions on rings sharing a larger amount of bonds with other rings
- Balance BN substitutions in each ring

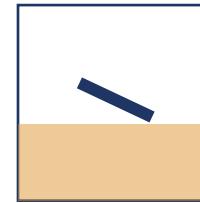
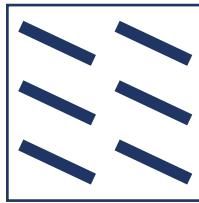
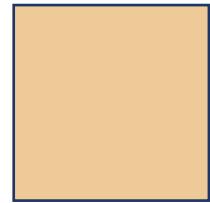
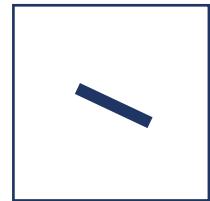
QA: Millions at once!

Not a single QM calculation required!

Future Steps

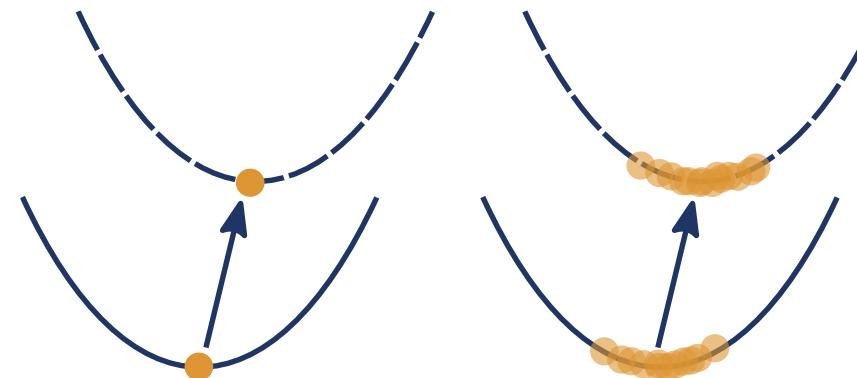
32

Include materials



also in collaboration with CINSaT members

Include ensembles

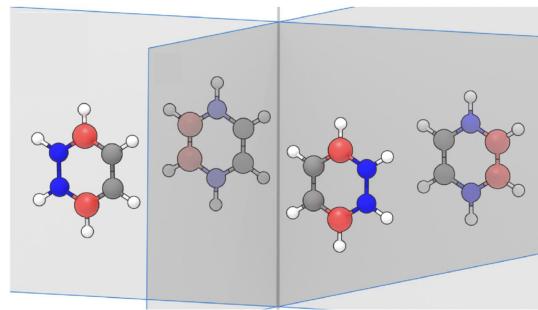
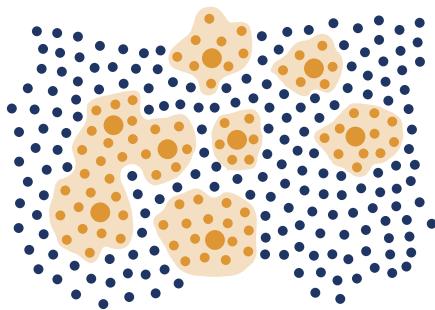


Method

- Basis sets and pseudopotentials need alchemy built-in
- Exploit pen&paper structure for fundamental aspects
- Push differentiable quantum chemistry
- Simplify use

Summary

33



- Efficient | Re-use knowledge, no one-by-one
- Symmetries | Reducing (“folding”) search space
- Constraints | Exclude whole regions
- Differentiable Chemistry | Arbitrary derivatives
- Representations | Better data efficiency

Thanks
Marco Bragato
Giorgio Domenichini
Emily Eikey
Stefan Heinen
Chasz Griego
Konstantin Karandashev
John Keith
Mario Krenn
Simon Krug
Dominik Lemm
Anatole von Lilienfeld
Alex Maldonado
Michael Sahre
Max Schwilk
Enrico Tapavicza
Jan Weinreich

Interpolate between molecular isoelectronic Hamiltonians

$$\hat{H}(\lambda) \equiv \lambda \hat{H}_t + (1 - \lambda) \hat{H}_r \quad \lambda \in [0, 1]$$

Taylor expansion around reference molecule

$$E_t = \sum_{n=0}^{\infty} \frac{1}{n!} \frac{\partial^n}{\partial \lambda^n} \left\langle \psi_{\lambda} \left| \hat{H}(\lambda) \right| \psi_{\lambda} \right\rangle \Big|_{\lambda=0} = E_r + \sum_{n=1}^{\infty} \frac{1}{n!} \left. \frac{\partial^n E(\lambda)}{\partial \lambda^n} \right|_{\lambda=0}$$

Hellmann-Feynman theorem

$$\partial_{\lambda} E = \left\langle \psi_{\lambda} \left| \hat{H}_t - \hat{H}_r \right| \psi_{\lambda} \right\rangle = \Delta E^{NN} + \int_{\Omega} d\mathbf{r} \underbrace{(v_t(\mathbf{r}) - v_r(\mathbf{r}))}_{\equiv \Delta v} \rho_{\lambda}(\mathbf{r})$$

Alchemical Perturbation Density Functional Theory (APDFT)

$$E_t = E_r + \Delta E^{\text{NN}} + \int_{\Omega} d\mathbf{r} \sum_{n=0}^{\infty} \frac{1}{(n+1)!} \left. \Delta v \frac{\partial^n \rho_{\lambda}(\mathbf{r})}{\partial \lambda^n} \right|_{\lambda=0}$$

$$\rho_t = \sum_{n=0}^{\infty} \frac{1}{n!} \left. \frac{\partial^n \rho}{\partial \lambda^n} \right|_{\lambda=0}$$

- Gives consistent energies, densities, forces, ...
- Uses the same derivatives for all predictions



ferchault/APDFT