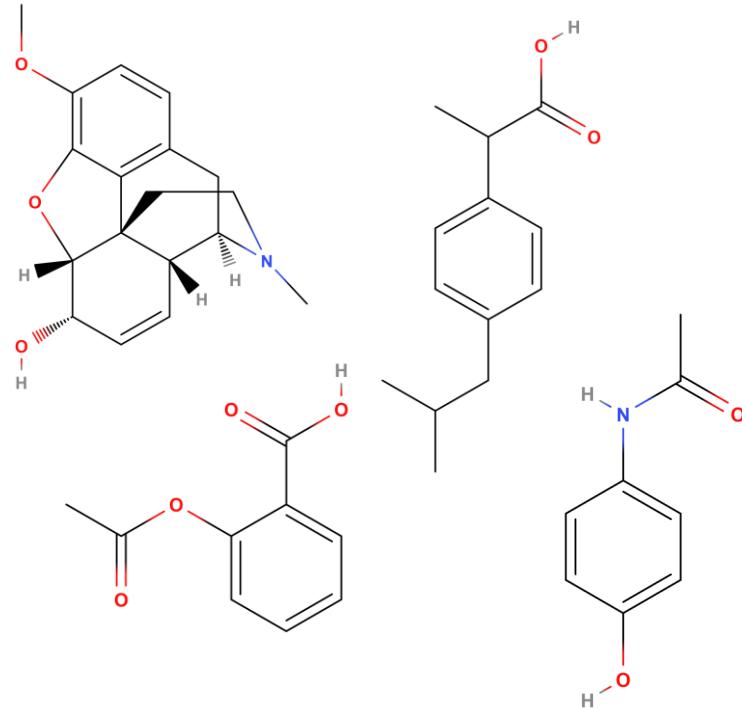


Design in Compound Space with Machine Learning and Quantum Alchemy

Guido Falk von Rudorff, University of Kassel

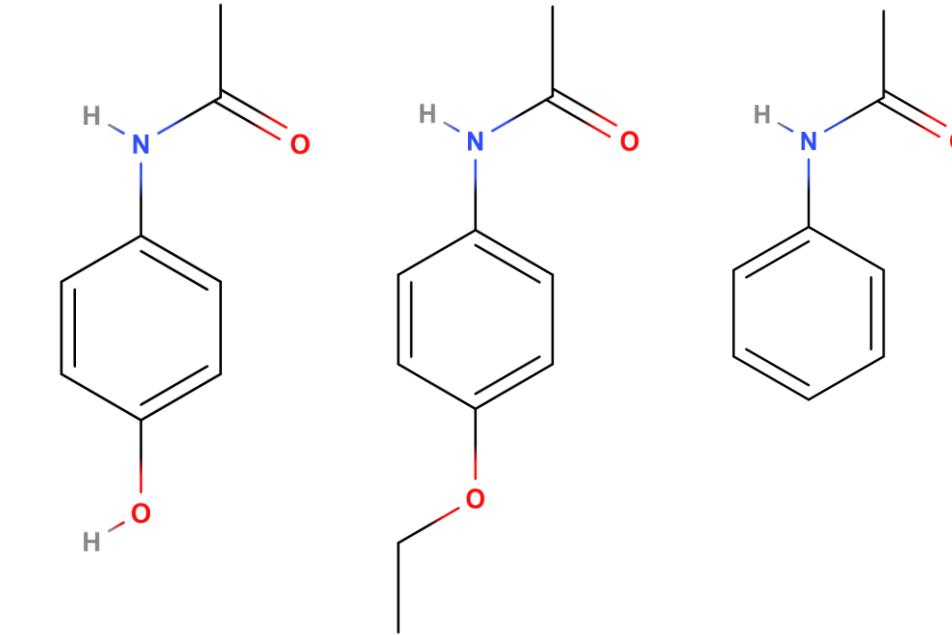
Introduction

2



Global Search Problem

Which class of compounds?

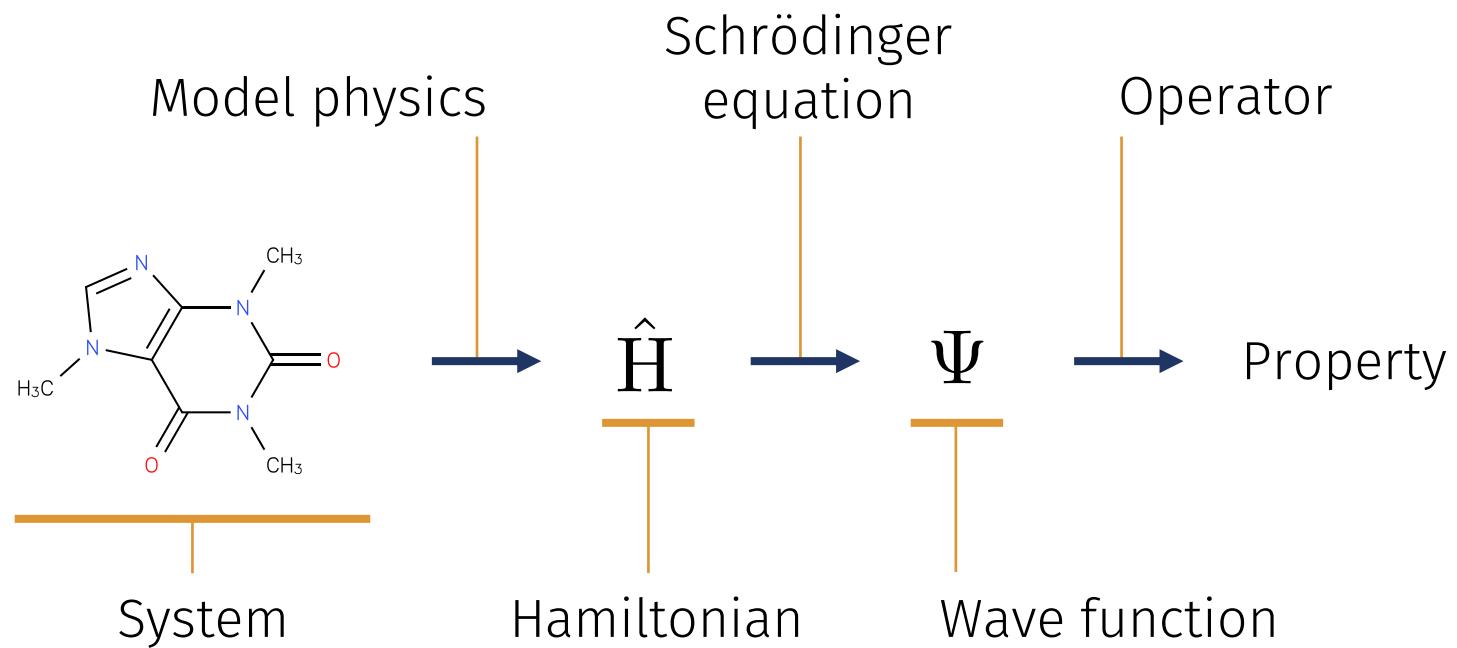


Local Search Problem

Which particular species within that class?

Solved?

3



Wave function

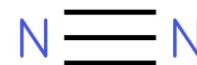
4

$$\Psi = \Psi(\mathbf{r}_1, \mathbf{r}_2, \mathbf{r}_3, \dots, \mathbf{r}_n)$$

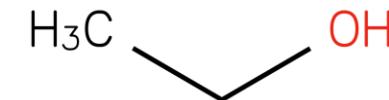
Methane



N₂



Ethanol



Solved by approximations in computational chemistry?



Chemistry at scale

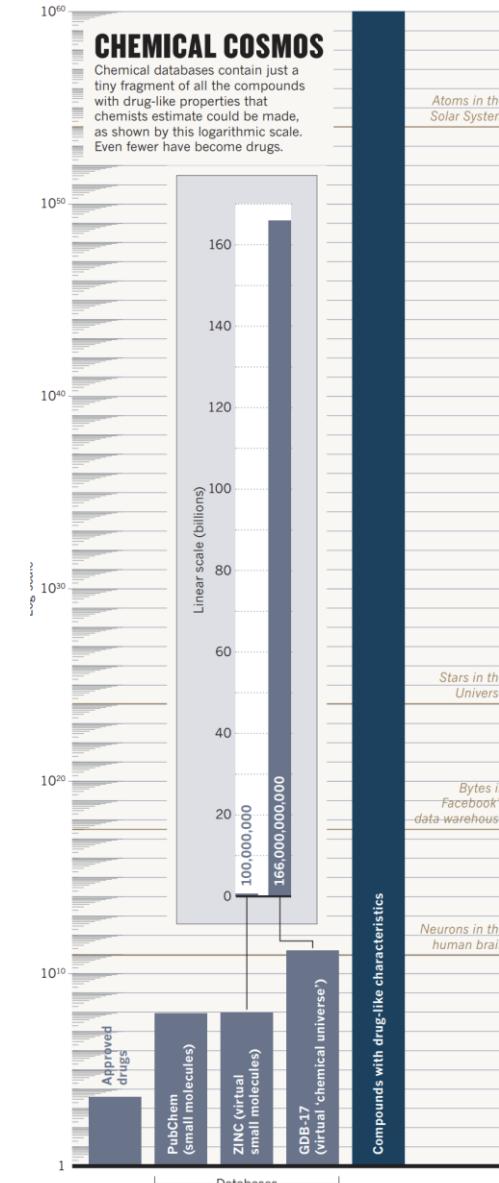
5

Commercial databases

- 164 million molecules
- 15k added daily

Scale

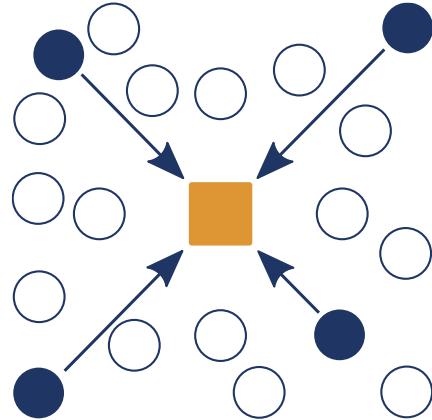
- One person: 1 million compounds/second
- 10 billion people on earth
- 10^{26} universe ages to go through



Introduction

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Machine Learning



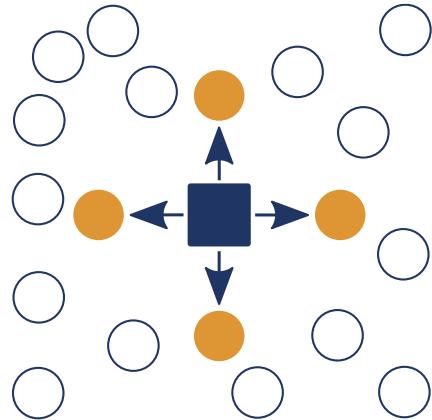
Foundations | Statistical modelling

Accuracy | Systematically improvable through data and training

Specialty | Universal, scale-bridging, data-driven approach

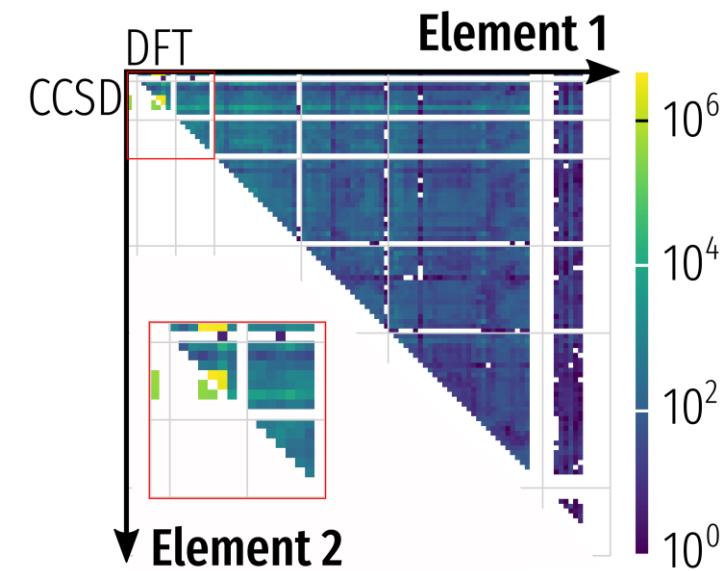
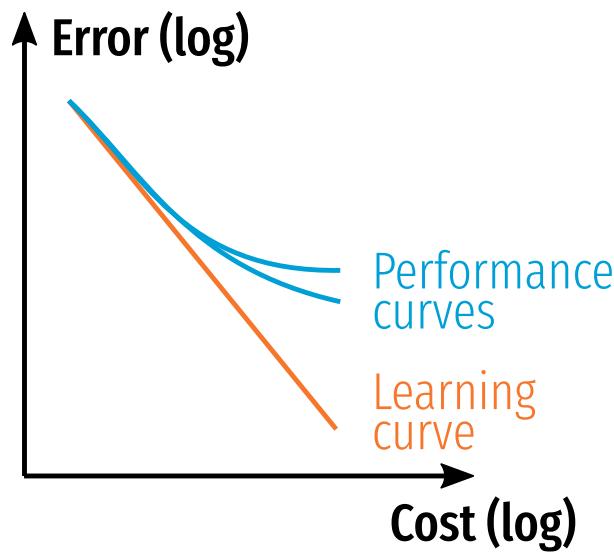
Limitation | Requires training data, no black box

Quantum Alchemy



Data availability

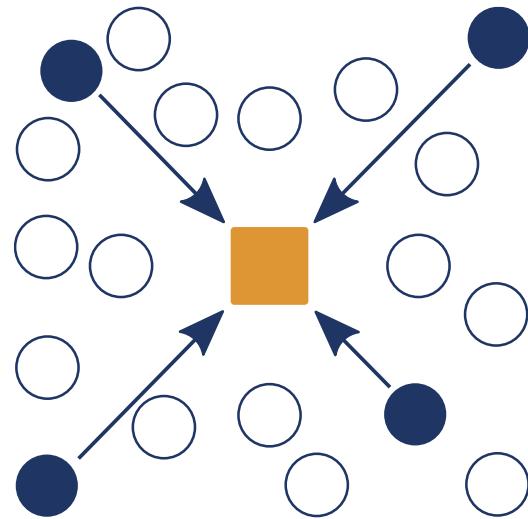
7



Introduction: Machine Learning

8

Machine Learning



Kernel-Ridge-Regression

- Efficient in the low-data regime (around 1k points)
- Ingredients
 - Representation
 - Similarity measure
 - Observed properties
- Training
 - Pairwise similarities
 - Model coefficients
- Predictions
 - Compare to training

 \mathbf{M}

$$k(\mathbf{M}_i, \mathbf{M}_j)$$

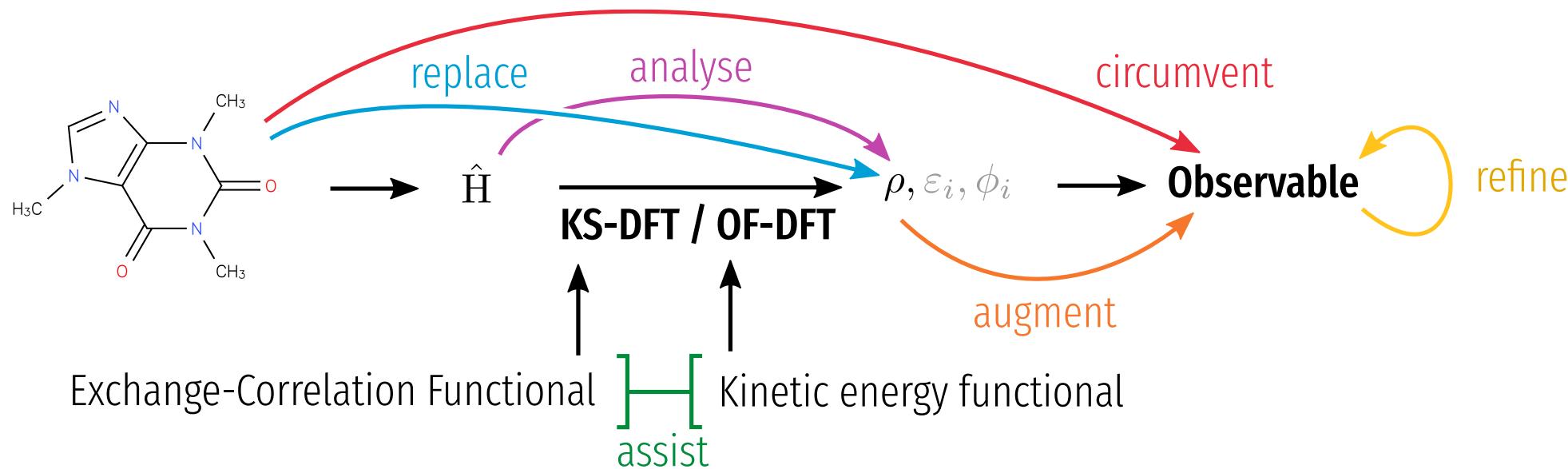
 \mathbf{y} \mathbf{K}

$$\boldsymbol{\alpha} = (\mathbf{K} + \lambda \mathbf{I})^{-1} \mathbf{y}$$

$$\tilde{q}(\mathbf{M}) = \sum_i \alpha_i k(\mathbf{M}, \mathbf{M}_i)$$

Strategies

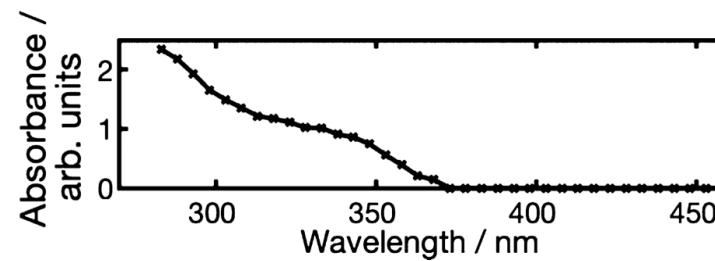
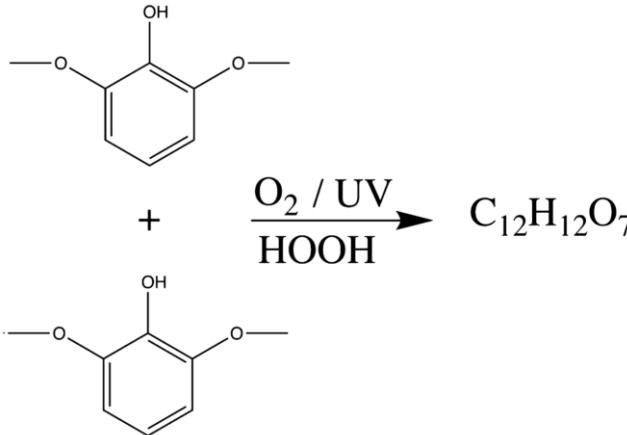
9



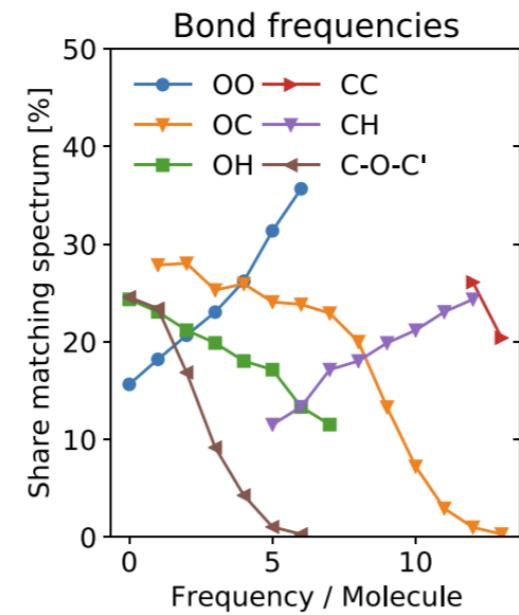
Identification of spectra

10

Experiment



Identified features



Search space

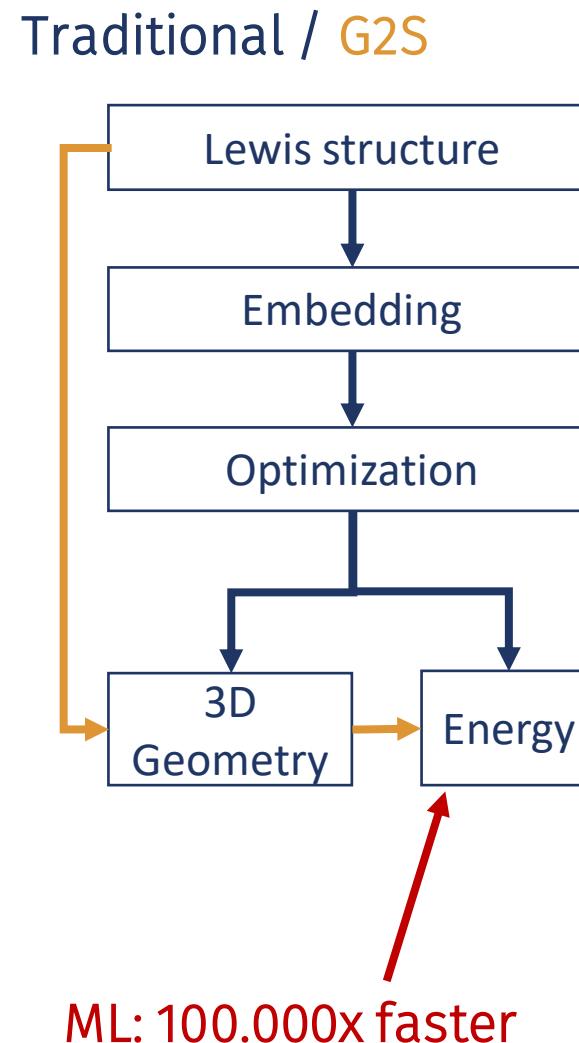
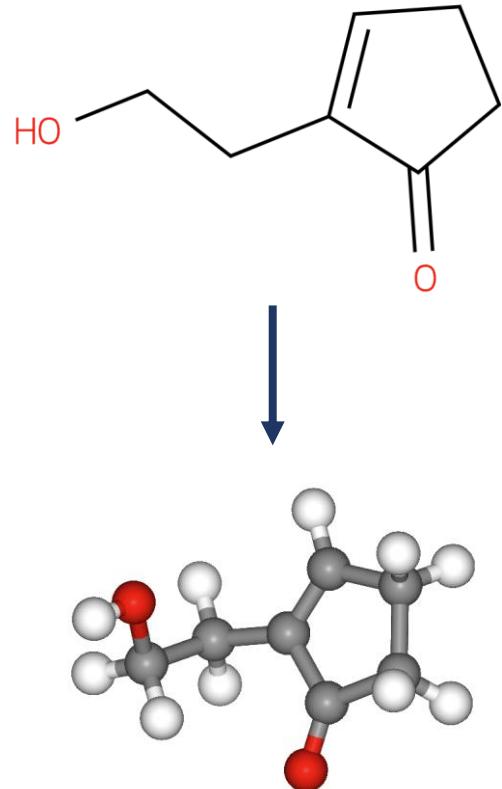
Molecular graphs: 264 M
Stable molecules: 123 M

Guide experiment

How many molecules are left and which feature to measure next?

Geometry learning

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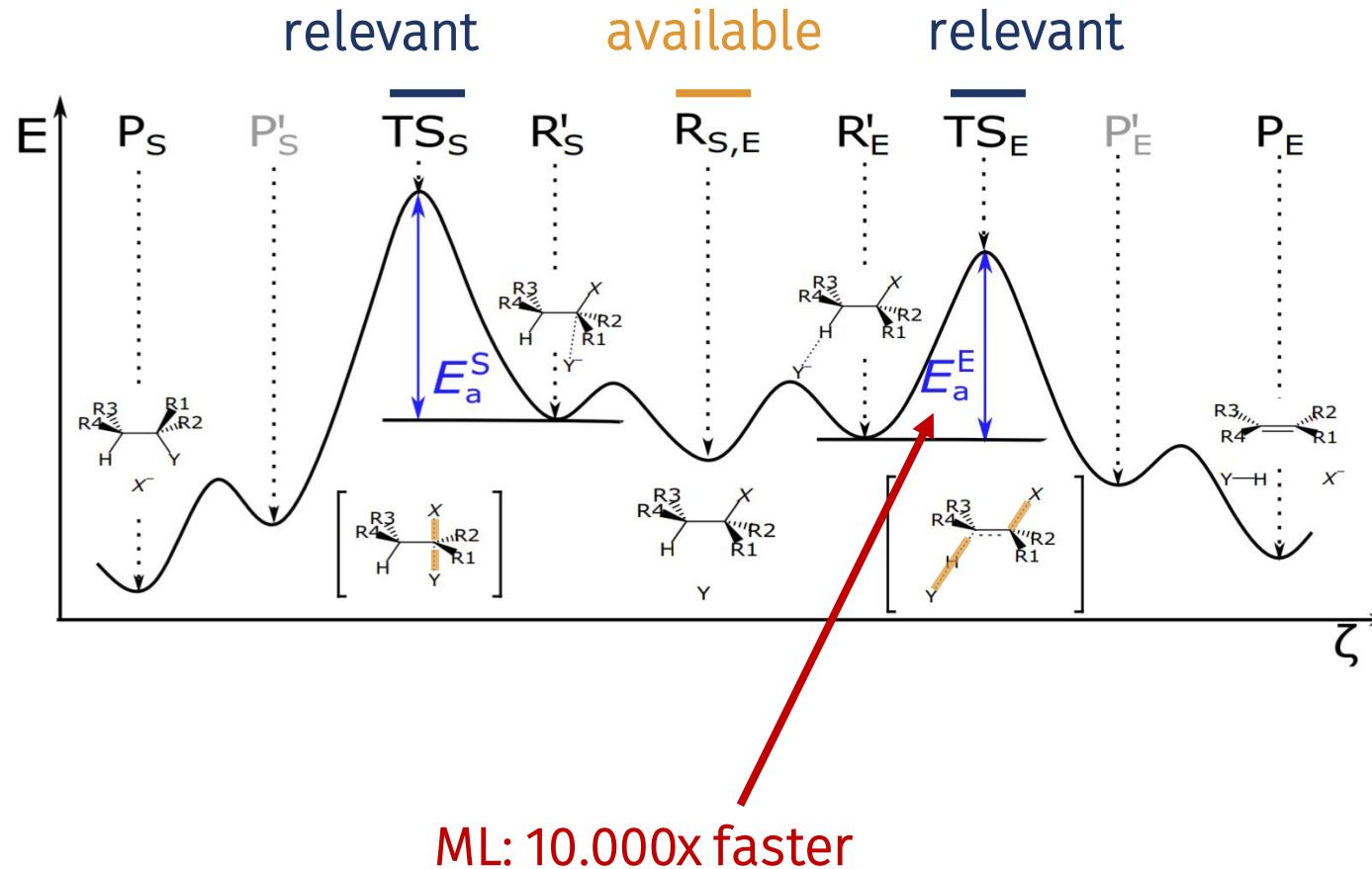


G2S

- Closer to DFT than common methods
 - Small molecules
- Applicable to complex chemical spaces
 - Transition state geometries
 - Carbenes
 - Elpasolite crystals

Reaction barrier learning

12

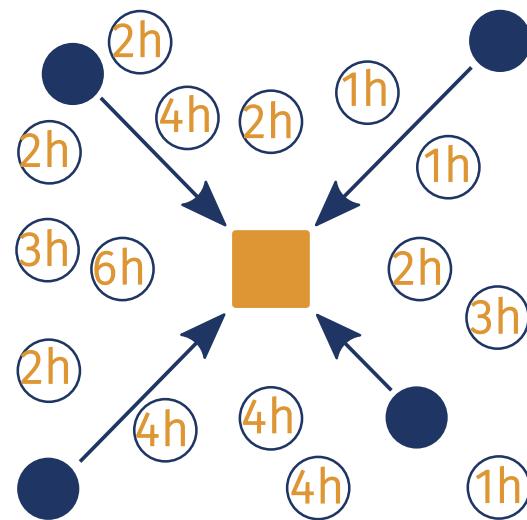
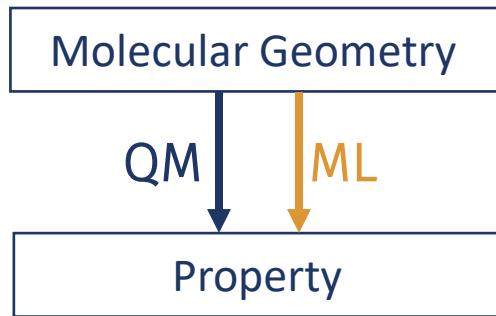


Competing reactions: $E2, S_N2$

- 4.5k reactions in one new dataset
- Learning activation energies from reactants only reaching 2.5 kcal/mol with 800 data points
- Learning geometries of transition states
 - direct
0.05 Angstrom for distances
 - G2S
0.45 Angstrom heavy-atom RMSD

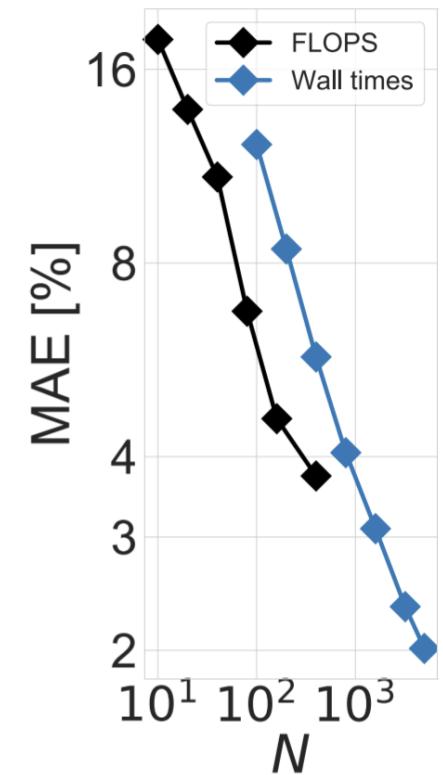
Computational cost

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Computational effort as molecular property

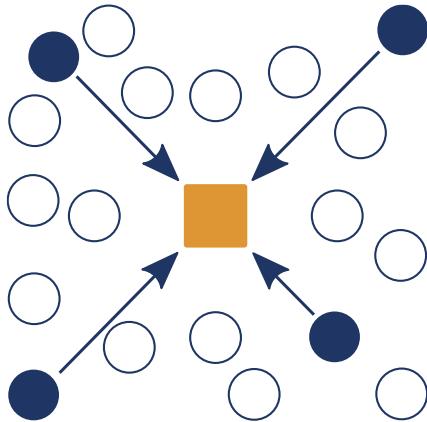
- Improves models
- Accuracy depends on problem
 - Single points: 2%
 - Transition state search: 25%
 - Geometry optimisations: 40%



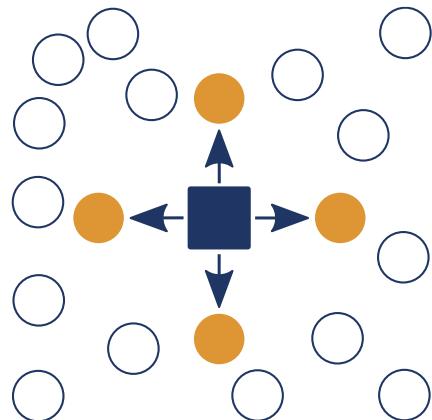
Introduction

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Machine Learning



Quantum Alchemy



Foundations | Perturbation theory

Accuracy | Systematically improvable through higher orders terms

Specialty | Combinatorial scaling with chemical diversity

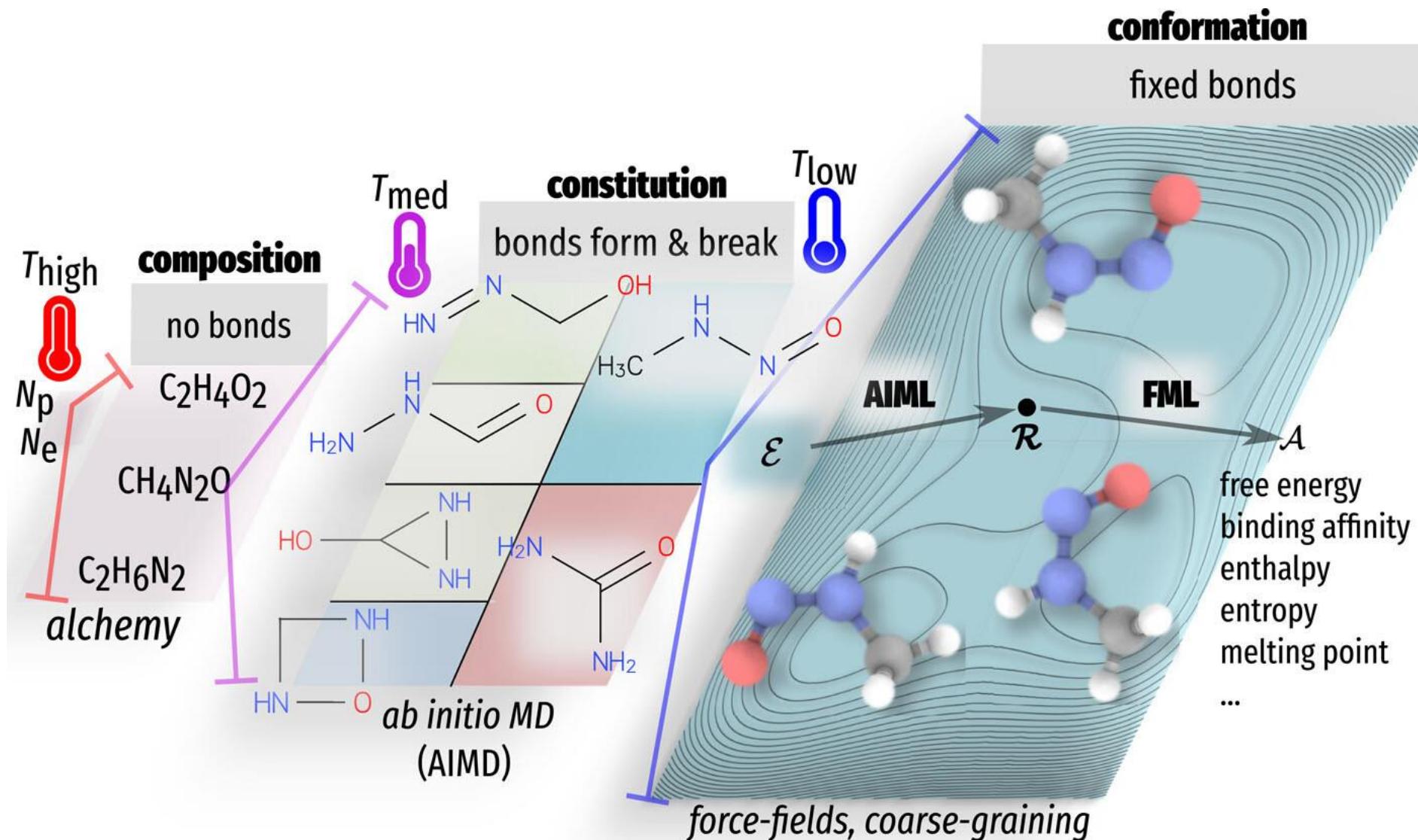
Limitation | Finite range in chemical space



Joseph Wright, 1771

Discrete points?

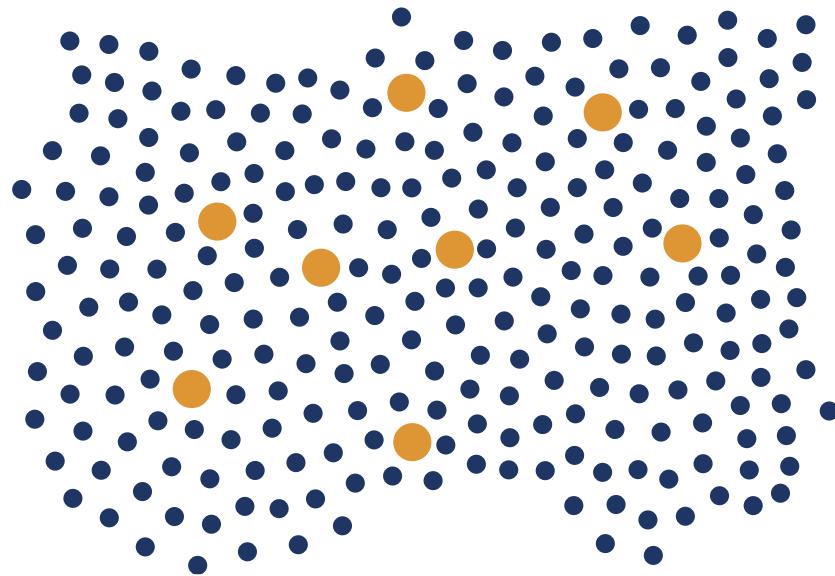
16



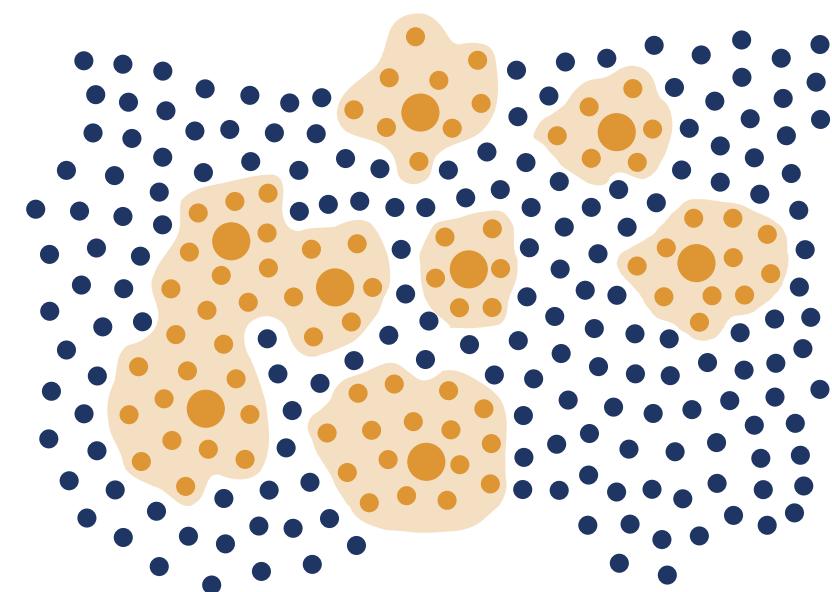
Introduction

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Without Perturbation



With Perturbation



Systems/Molecules

- Any
- Known
- Approximated

Perspective shift

Few highly accurate calculations
instead of many intermediate ones

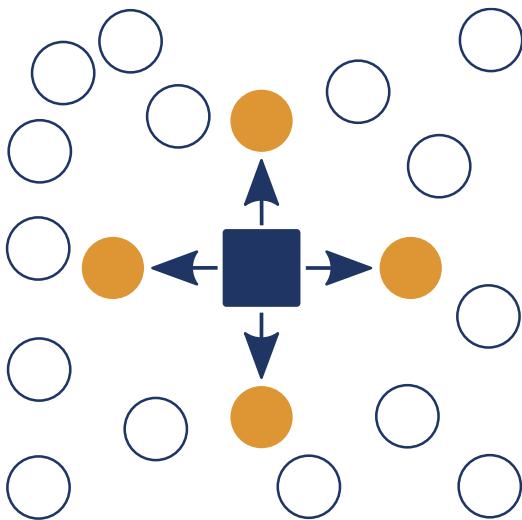
$$\hat{H} = \hat{H}(Z_i, \mathbf{R}_i, N_e, \sigma)$$

4N 1D, close to $\sum_i Z_i$ 1D

Quantum Alchemy

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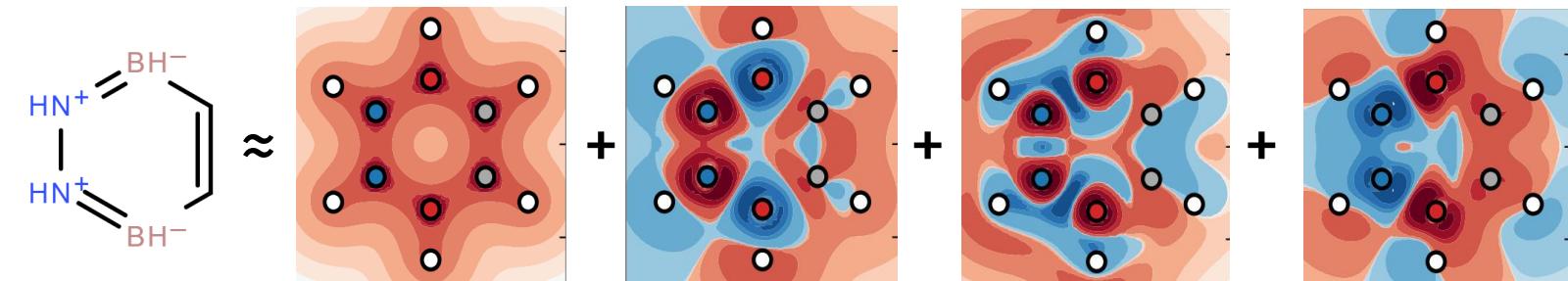
Quantum Alchemy



Taylor expansion

- Energy function of
 - Geometry
 - Nuclear charges
- Idea: obtain dominant leading derivatives, predict many systems

Forces, Vibrations
Alchemical changes



E. B. Wilson, *J. Chem. Phys.* 1962.

GFvR, O. A. von Lilienfeld, *Phys. Rev. Res.*, 2020.

Quantum Alchemy

20

Interpolate between molecular isoelectronic Hamiltonians

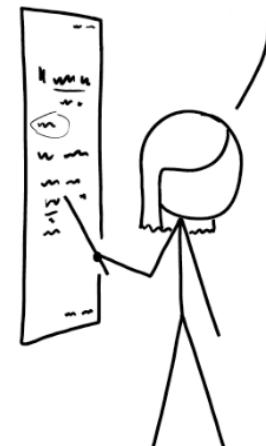
$$\hat{H}(\lambda) \equiv \lambda \hat{H}_t + (1 - \lambda) \hat{H}_r$$

Taylor expansion around reference molecule

$$E_t = E_r + \Delta E^{\text{NN}} + \int_{\Omega} d\mathbf{r} \sum_{n=0}^{\infty} \frac{1}{(n+1)!} \left. \Delta v \frac{\partial^n \rho_{\lambda}(\mathbf{r})}{\partial \lambda^n} \right|_{\lambda=0}$$

- Gives consistent energies, densities, forces, ...
- Uses the same derivatives for all predictions

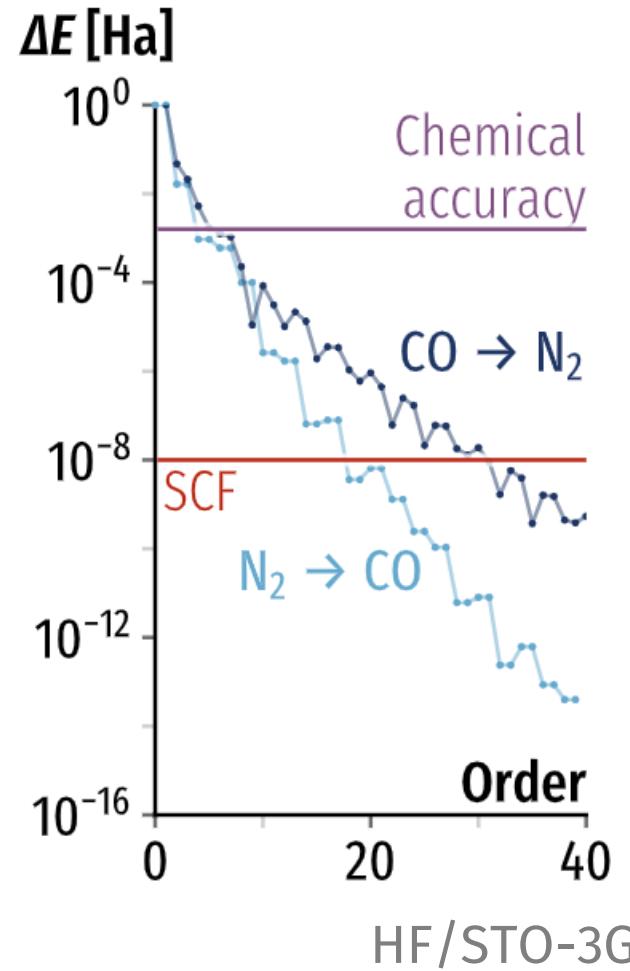
AT THIS POINT, YOU'RE PROBABLY
THINKING, "I LOVE THIS EQUATION
AND WISH IT WOULD NEVER END!"
WELL, GOOD NEWS!



xkcd.com/2605

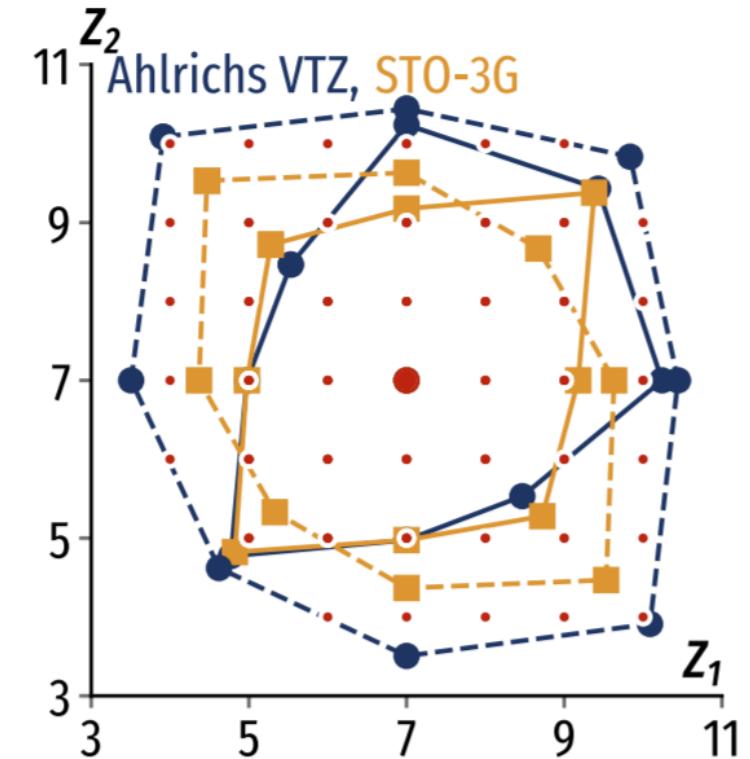
Convergence

21



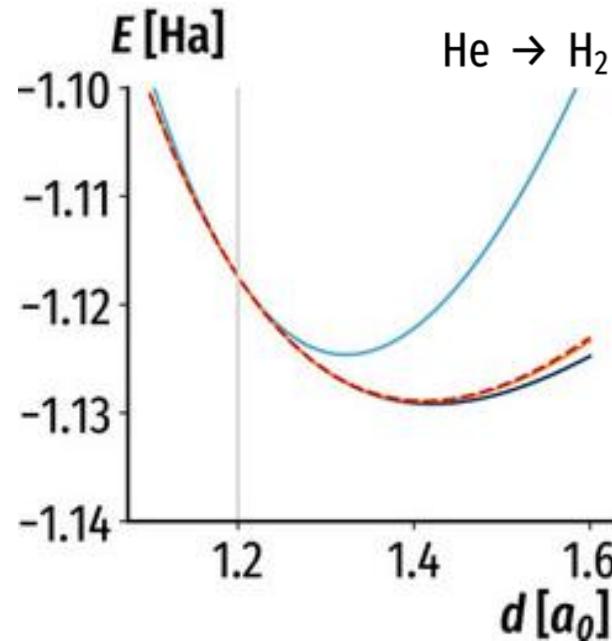
Taylor expansion

- First terms accurate enough
 - Truncate early
- Converges to the right value
- Large convergence radius
- Scales with chemical space



Geometry relaxation

22



Taylor expansion

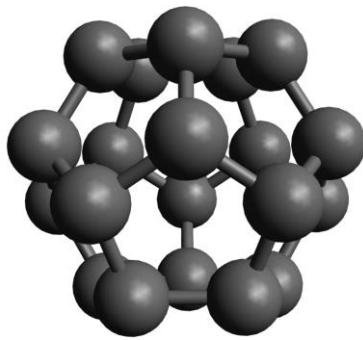
- Large changes still converge (more slowly)
- Geometric response can be recovered

Covalent interactions

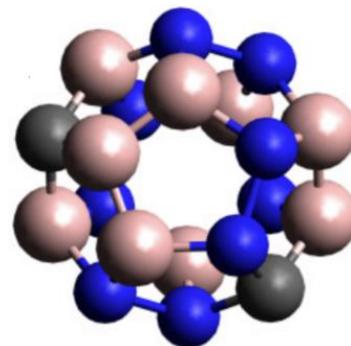
23

Scaling with chemical space

- 1 derivative for second order
- 5 derivatives for third order

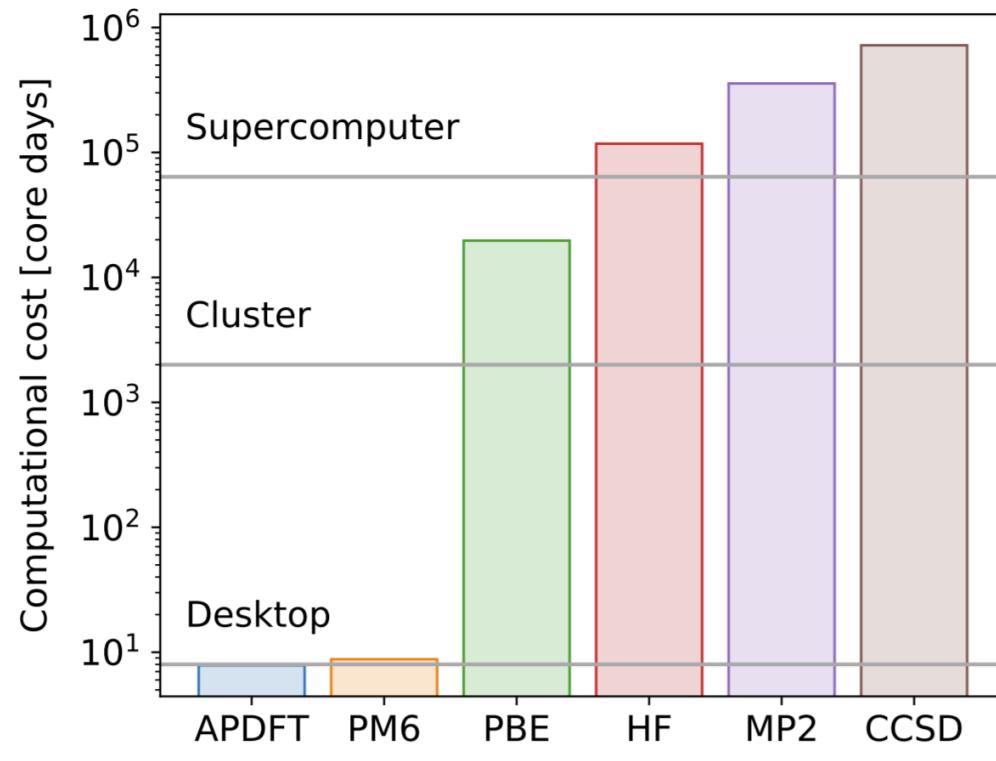


C_{20}



$3.1 \cdot 10^6$
targets

QA: 80.000x faster

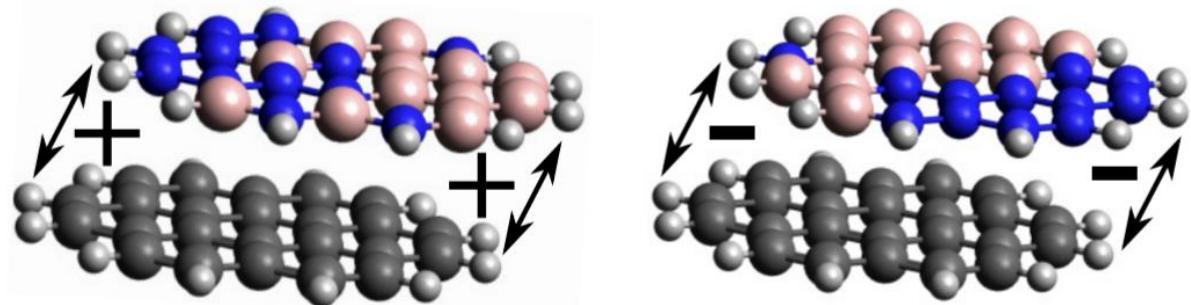


Non-covalent interactions

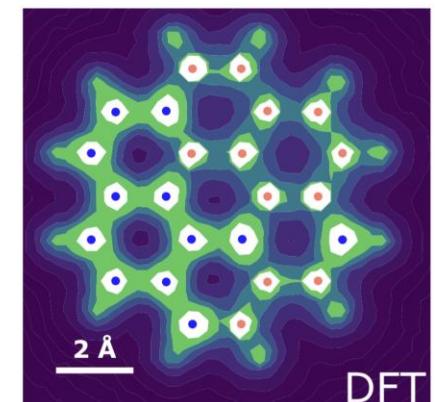
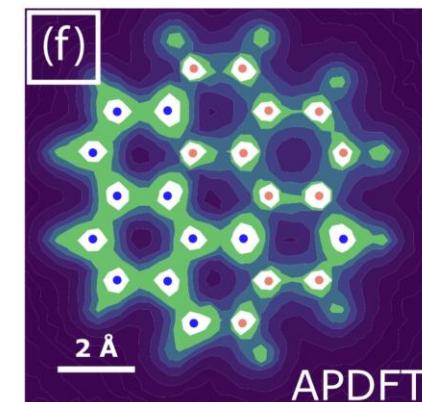
24

BN-doped coronene dimer

- Identify most/least attractive doping pattern
- Design case



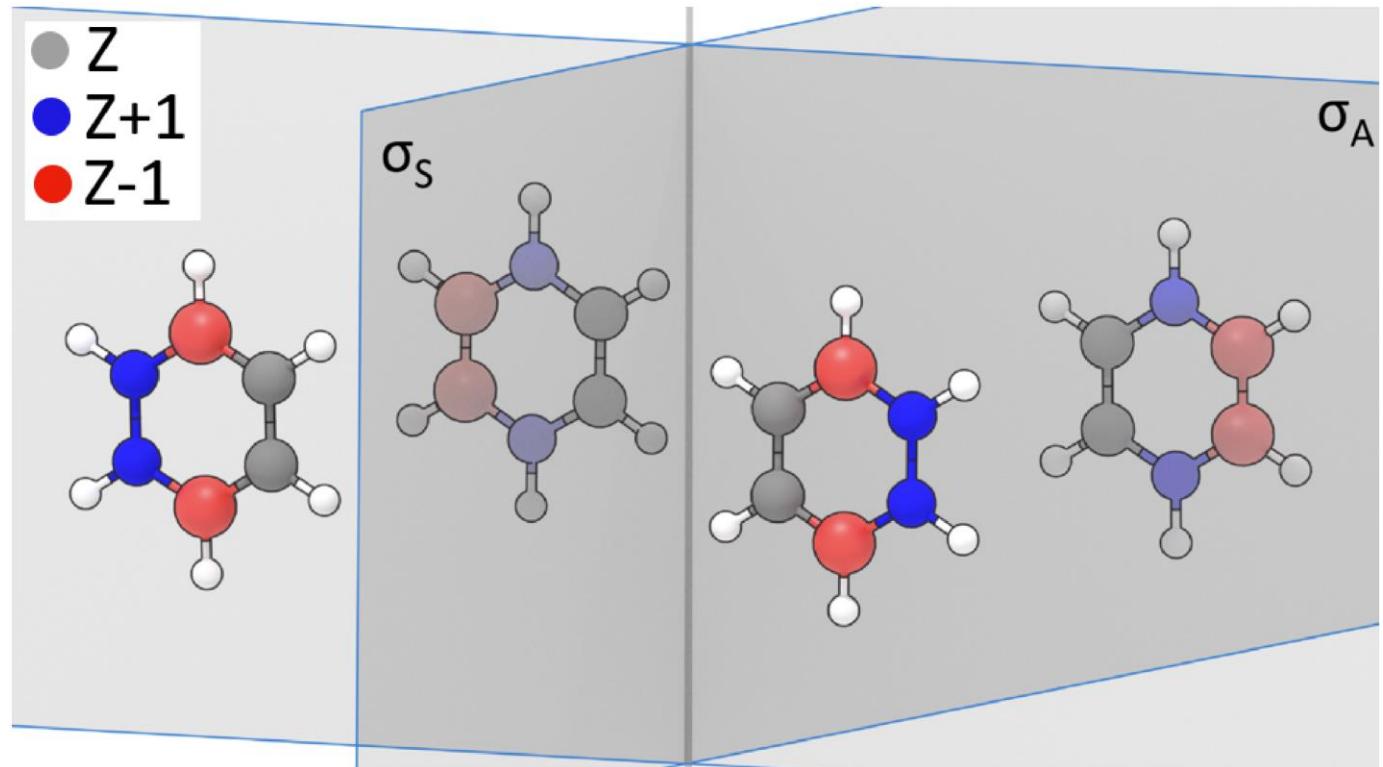
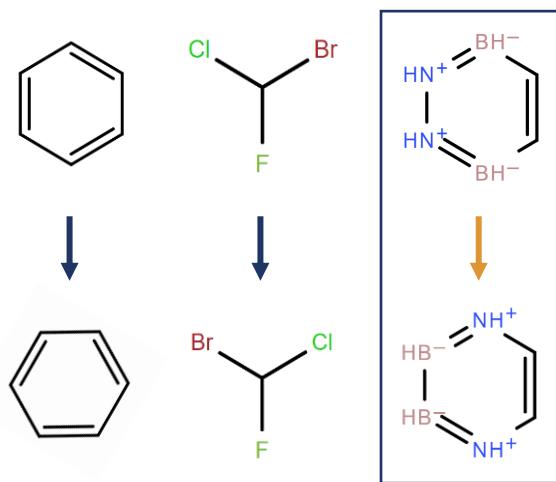
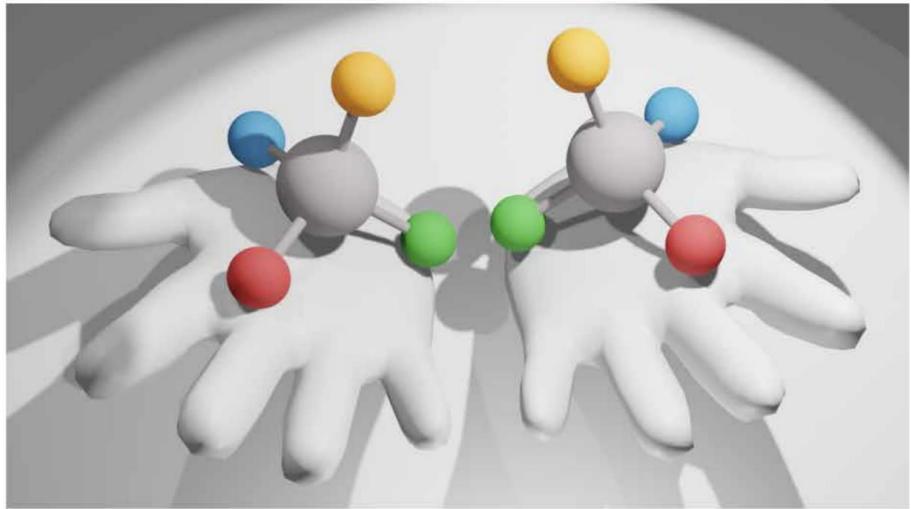
QA: 20.000x faster



$2.8 \cdot 10^{10}$ targets

Alchemical enantiomers

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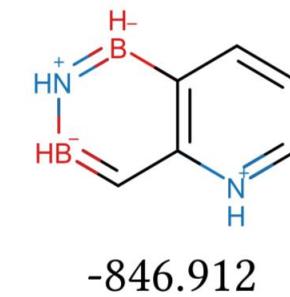
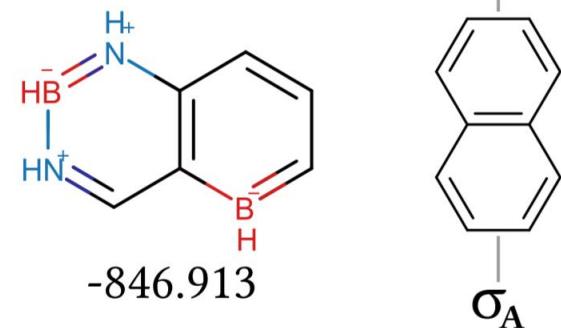


Alchemical enantiomers

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Fundamentally new symmetry

Electronic energy only



Bond energy rules

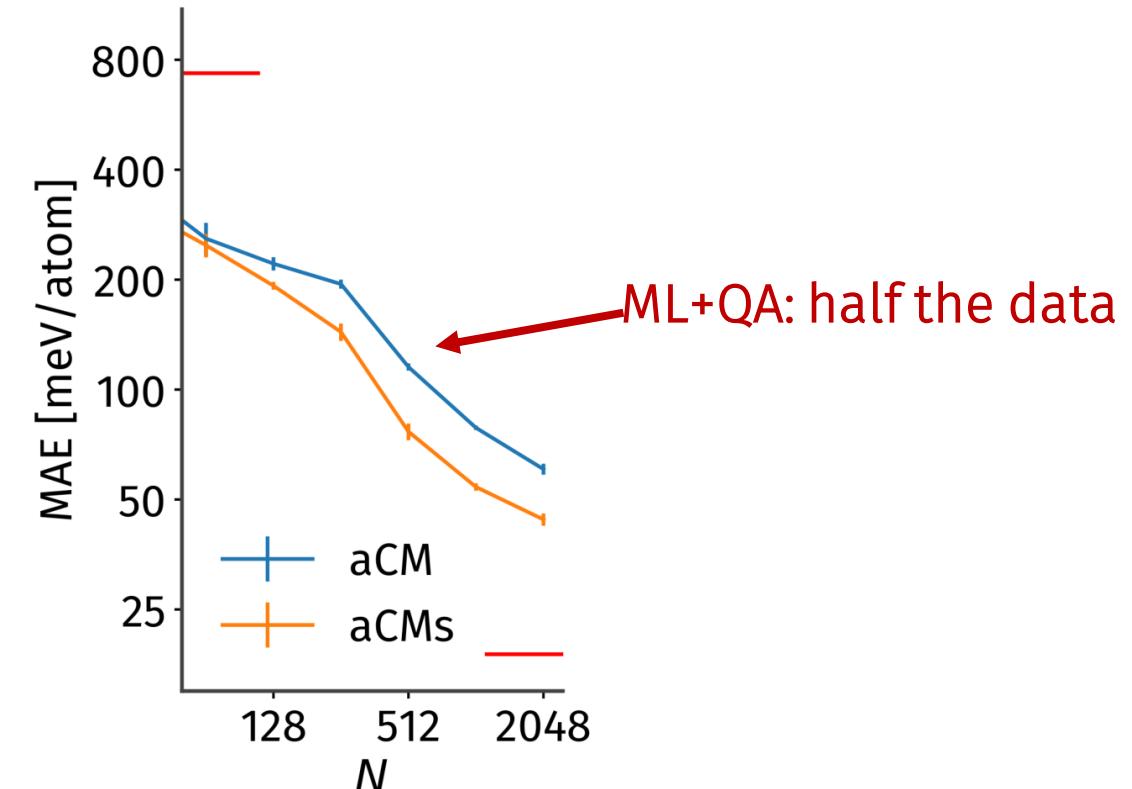
Consecutive Elements

Q R S

B C N

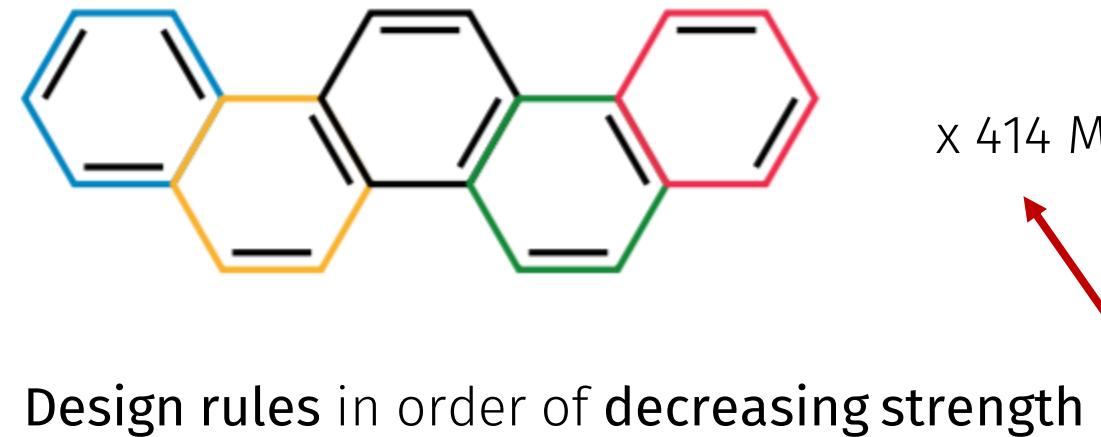
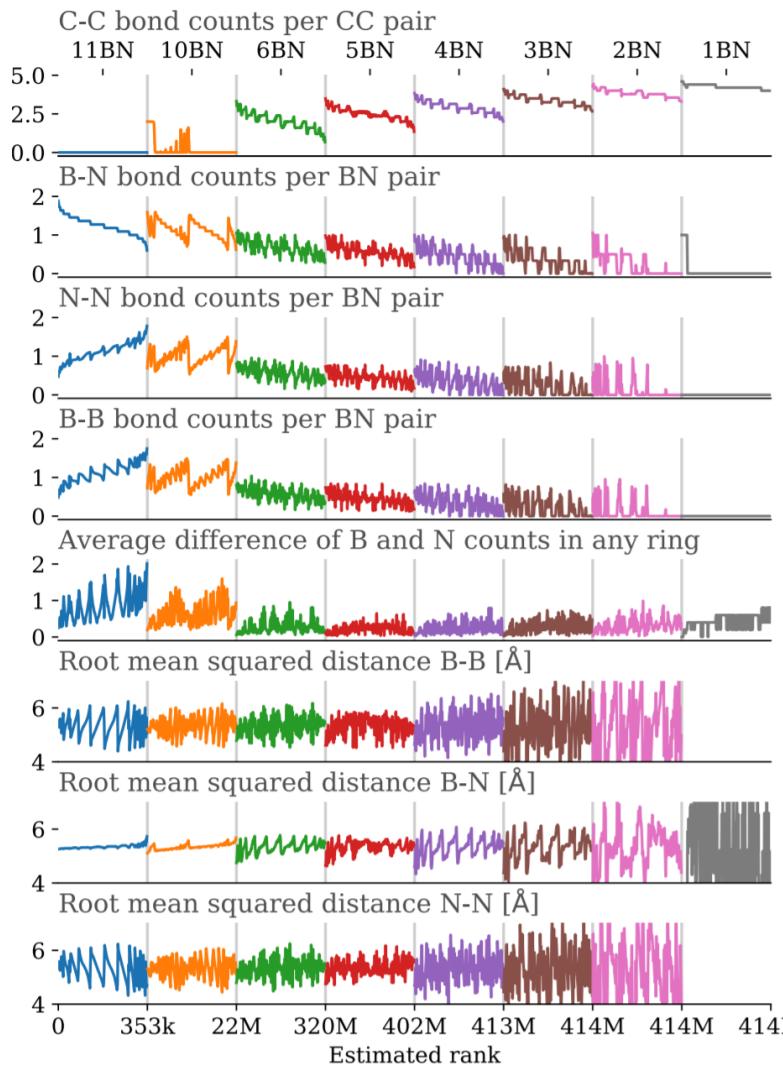
$$E_{QR} \approx E_{SR} + 0.5(E_{QQ} - E_{SS})$$

Speed up machine learning



Alchemical enantiomers

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Design rules in order of decreasing strength

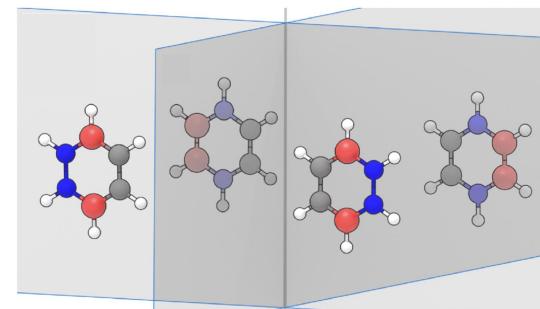
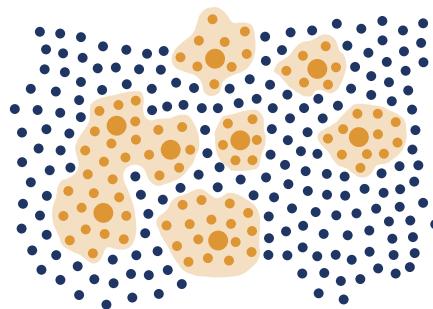
- Add BN pairs
- Maximize CC bonds
- Substitute sites shared between rings
- Maximize BN bonds
- Avoid N substitutions on rings sharing a larger amount of bonds with other rings
- Balance BN substitutions in each ring

QA: Millions at once!

Not a single QM calculation required!

Summary

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Efficient | Re-use knowledge, no one-by-one

Symmetries | Reducing (“folding”) search space

Constraints? | Exclude regions of interest

Differentiable Chemistry? | Arbitrary derivatives

Representations? | Better data efficiency

Ensembles? | Derivatives on dynamic observables

Thanks

Marco Bragato
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Konstantin Karandashev
John Keith
Mario Krenn
Simon Krug
Dominik Lemm
Anatole von Lilienfeld
Alex Maldonado
Michael Sahre
Max Schwilk
Enrico Tapavicza
Jan Weinreich

Interpolate between molecular isoelectronic Hamiltonians

$$\hat{H}(\lambda) \equiv \lambda \hat{H}_t + (1 - \lambda) \hat{H}_r \quad \lambda \in [0, 1]$$

Taylor expansion around reference molecule

$$E_t = \sum_{n=0}^{\infty} \frac{1}{n!} \frac{\partial^n}{\partial \lambda^n} \left\langle \psi_{\lambda} \left| \hat{H}(\lambda) \right| \psi_{\lambda} \right\rangle \Big|_{\lambda=0} = E_r + \sum_{n=1}^{\infty} \frac{1}{n!} \left. \frac{\partial^n E(\lambda)}{\partial \lambda^n} \right|_{\lambda=0}$$

Hellmann-Feynman theorem

$$\partial_{\lambda} E = \left\langle \psi_{\lambda} \left| \hat{H}_t - \hat{H}_r \right| \psi_{\lambda} \right\rangle = \Delta E^{NN} + \int_{\Omega} d\mathbf{r} \underbrace{(v_t(\mathbf{r}) - v_r(\mathbf{r}))}_{\equiv \Delta v} \rho_{\lambda}(\mathbf{r})$$

Alchemical Perturbation Density Functional Theory (APDFT)

$$E_t = E_r + \Delta E^{\text{NN}} + \int_{\Omega} d\mathbf{r} \sum_{n=0}^{\infty} \frac{1}{(n+1)!} \left. \Delta v \frac{\partial^n \rho_{\lambda}(\mathbf{r})}{\partial \lambda^n} \right|_{\lambda=0}$$

$$\rho_t = \sum_{n=0}^{\infty} \frac{1}{n!} \left. \frac{\partial^n \rho}{\partial \lambda^n} \right|_{\lambda=0}$$

- Gives consistent energies, densities, forces, ...
- Uses the same derivatives for all predictions

